An exciting research challenge in supramolecular chemistry is to design, synthesize, and characterize nano-sized architectures with applications in biology, chemistry, and materials science.¹ Predicting and designing non-covalently bound supramolecular complexes and assemblies is difficult because of the weakness of the interactions involved, thus the resulting superstructure is often a compromise between the geometrical constraints of the building blocks and the competing weak intermolecular interactions.²

Our research interest has been focused on the studies of weak non-covalent intermolecular, viz. supramolecular interactions as the driving force in self-assembly and molecular recognition, especially in the solid state by single crystal X-ray diffraction. The lecture will highlight some of our recent studies on hydrogen³ and halogen4 bonded systems, $\pi - \pi^5$, CH...anion6, and anion... π^7 interactions and metal ion coordination⁸ in molecular self-assembly and molecular recognition in various systems such as resorcinarenes, ditopic receptors, rotaxanes, M₄L₆ tetrahedra, M₈L₆ cube, spheres, knots, etc..

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Seminar #2.4 February 26, 2021, 11:30 am

Supramolecular crystallography (interplay of intermolecular interactions in the solid-state)

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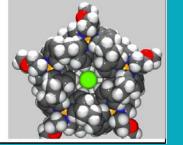


Fig. 1. The X-ray structure of the pentafoil knot. [8b]

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