

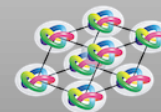


Supramolecular Upcycling of Protein Materials

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Workshop «Materiali da Proteine – Bioeconomy Day 2nd Ed.»
Politecnico di Milano, Italy - October 28, 2025





The Halogens...

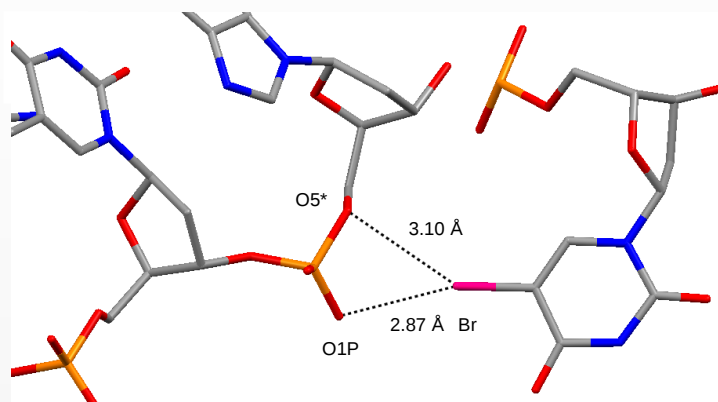


The %'s of halogens in the rocks of Earth's crust are 0.06 F, 0.031 Cl, 0.00016 Br, and 0.00003 I.
Seawater contains 536 mM chloride, 800 μ M bromide, and 0.5 μ M iodide anions.

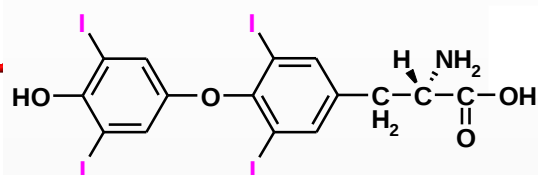
halogen	molecule	structure	model	$d(X-X)$ / pm (gas phase)	$d(X-X)$ / pm (solid phase)
fluorine	F ₂	$F-F$ 143 pm		143	149
chlorine	Cl ₂	$Cl-Cl$ 199 pm		199	198
bromine	Br ₂	$Br-Br$ 228 pm		228	227
iodine	I ₂	$I-I$ 266 pm		266	272

19 F 9
35 Cl 17
80 Br 35
127 I 53

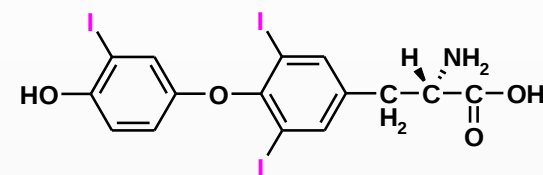
More than 4500 naturally occurring organohalogen compounds are known. Most natural organohalogens contain Cl (2300) or Br (2100), but a significant number contain I (120) or F (30).



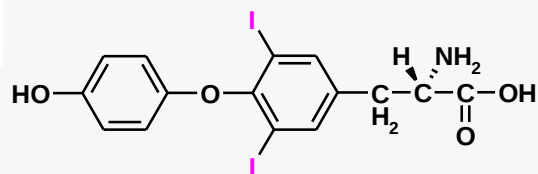
The short halogen bond between the bromine atom of a brominated uracil residue and the phosphate oxygens in a four-stranded DNA junction.



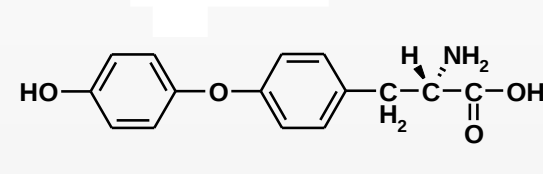
L-thyroxine (T4)
(71 ± 6%)



3,3',5-triiodo-L-thyronine (T3)
(62 ± 4%)



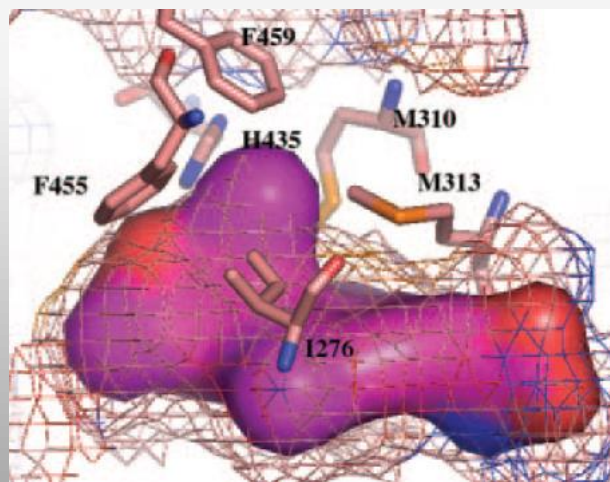
3,5-diiodo-L-thyronine (T2)
(15 ± 2%)



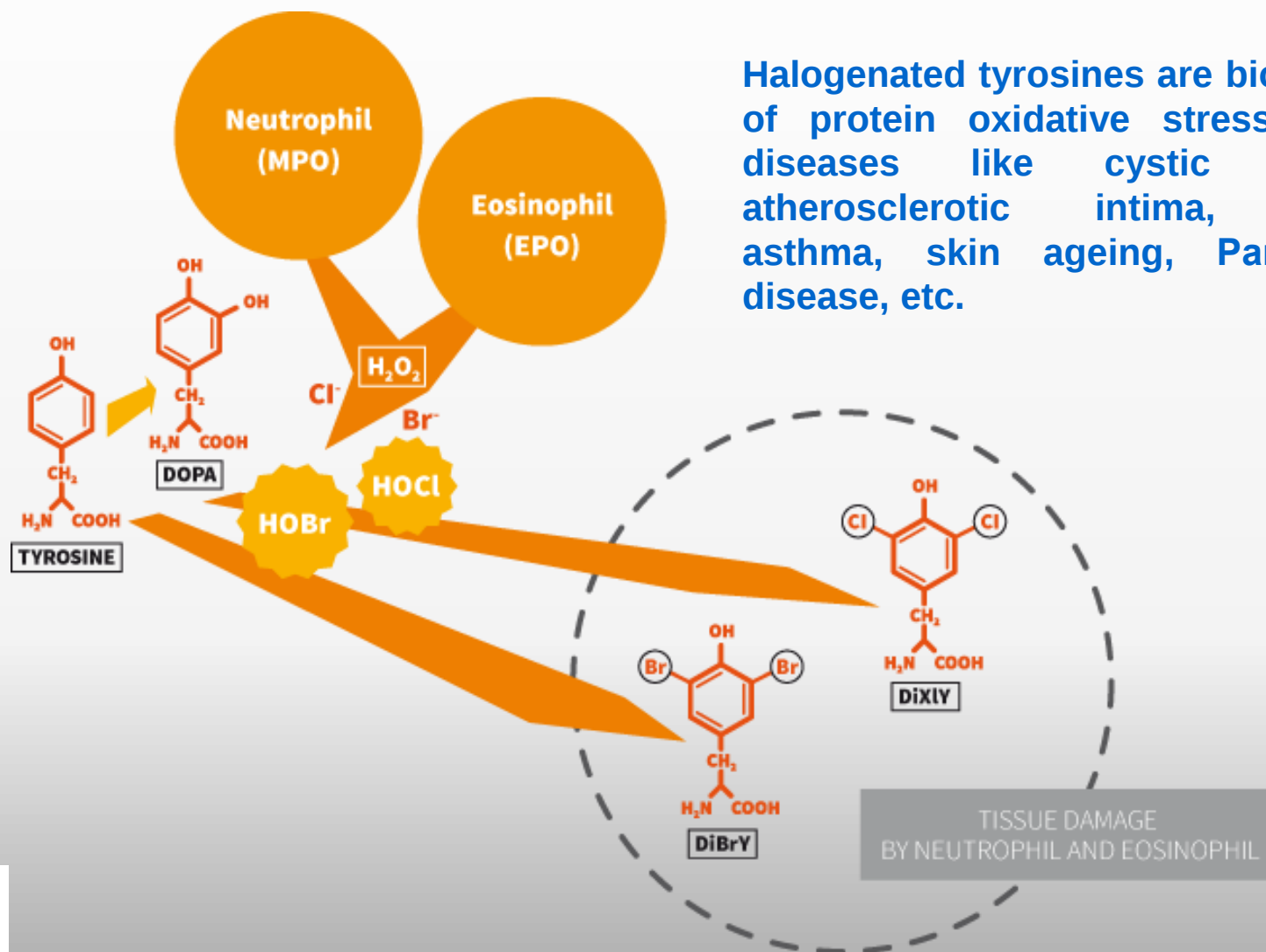
L-thyronine (T0)
(3 ± 1%)

Molecular structures of thyroxine derivatives and the binding activities of the RNA aptamer.

A. R. Voth, F. A. Hays, P. S. Ho *PNAS* 2007, 104, 6188;
D. Levésque, J.-D. Beaudoin, S. Roy, J.-P. Perreault *Biochem. J.* 2007, 403, 129; B. Sandler, P. Webb, J. W. Apriletti, B. R. Huber, M. Togashi, S. T. Cunha Lima, S. Juric, S. Nilsson, R. Wagner, R. J. Fletterick, J. D. Baxter *J. Biol. Chem.* 2004, 279, 55801.



Pocket of the TR-T4 complex.



Halogenated tyrosines are biomarkers of protein oxidative stress related diseases like cystic fibrosis, atherosclerotic intima, sepsis, asthma, skin ageing, Parkinson's disease, etc.

Invertebrates modify the mechanical properties and structures of their biomaterials either via calcification or by the introduction of heavy elements (Br, Zn, Cl, I).

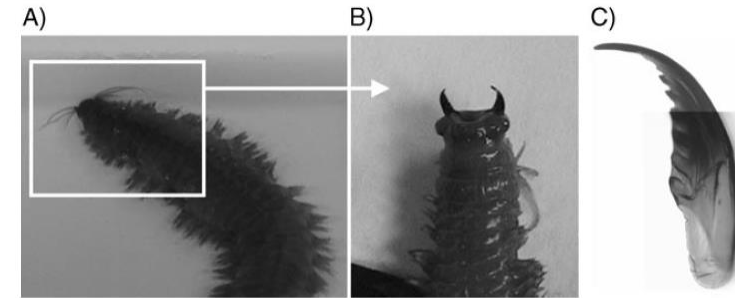
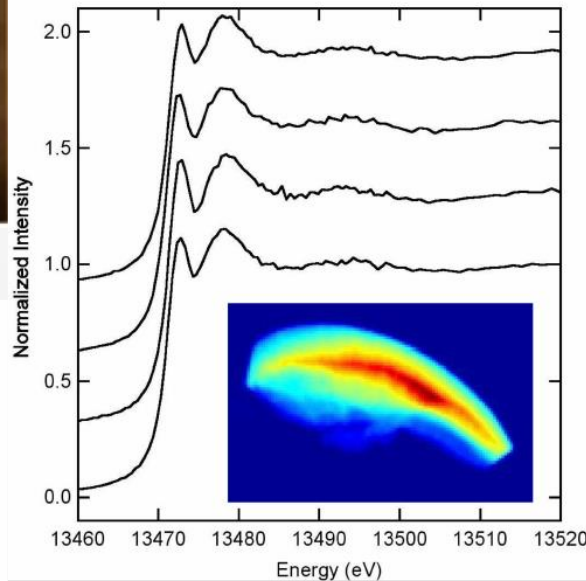
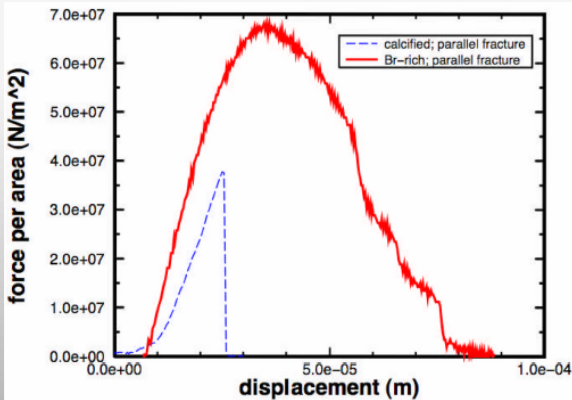
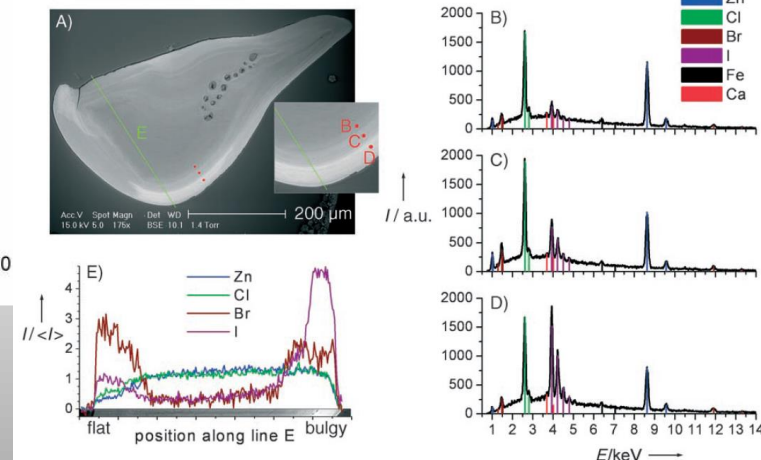


Figure 1. *N. virens* worm A) shown exploring, and B) with everted proboscis; the jaws are shown from the ventral side. C) A jaw extracted from the animal.



Chem. Eur. J. 2001, 7, 2511 (Concept Article, 978 cit.)

Acc. Chem. Res. 2005, 38, 386 (1764 citations)

Angew. Chem. Int. Ed. 2008, 47, 6114 (Minireview, 1419 citations)

Science 2008, 321, 918

CrystEngComm 2009, 11, 1187 (Highlight)

Nature Chem. 2011, 3, 260

Chem. Eng. News 2012, 90, 36 (27 August)

Pure Appl. Chem. 2013, 85, 1711
(IUPAC project no. 2009-032-1-100, 1598 cit.)

Chem. Rev. 2016, 116, 2478 (3007 citations)

SCIENCE & TECHNOLOGY

HALOGEN BONDING DEFINED

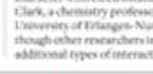
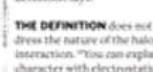
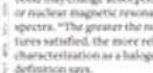
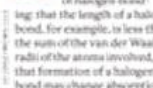
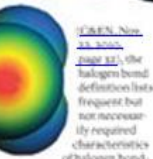
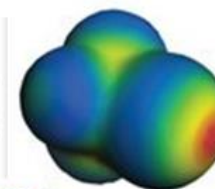
IUPAC effort sets parameters for **STRUCTURAL INTERACTION** seen in materials science and biology

HALOGEN BONDING, akin to hydrogen bonding, is a noncovalent interaction between a positive region on a halogen atom and a negative site, such as a lone pair on a nitrogen or oxygen. Following on the heels of a new definition for hydrogen bonding, an International Union of Pure & Applied Chemistry (IUPAC) committee has now proposed a definition for halogen bonding. The definition will promote recognition of halogen bonding as a key structural element in materials science and biology, scientists working in the field say.

The proposed definition says that a halogen bond is a net attractive interaction between an electrophilic region of a halogen atom on a molecule or molecular fragment and a nucleophilic region of another molecule or molecular fragment. The electrophilic region arises on the halogen because when a halogen makes a covalent bond in a molecule, the electron distribution around the atom shifts a bit toward the bond. That leaves an area of diminished electron density opposite the bond—an area christened as the σ -hole.

The σ -hole results in a region of positive electrostatic potential surrounded by an equatorial band of negative electrostatic potential. The geometry of the potentials around the halogen atom means that, in contrast to hydrogen bonds, halogen bonds are directional, occurring only in line with the halogen's molecular bond. Even fluorine can form halogen bonds, if it is covalently bonded to sufficiently electron-withdrawing groups, such as in F_2COE .

Like the hydrogen bond definition



that halogen bonds have some covalent character.

The reason for developing the definition is growing recognition of the roles halogen bonds play in the structure of materials, from inorganic to biological, say IUPAC committee chairs Pierangelo Metrangola and Giuseppe Resnati, chemistry professors at the Politecnico di Milano, in Italy. Metrangola and Resnati have explored halogen-bonding effects in liquid crystals, ion recognition and binding in supramolecular chemistry, and light-induced surface patterning of supramolecular polymers.

The area in which halogen bonding is poised to make big changes is in biochemistry. Current modeling software treats halogens as having all-negative electrostatic potential, eliminating the possibility of predicting halogen bonds. Yet many natural protein inhibitors involve halogens, notes F. Shing Ho, a biochemistry professor at Colorado State University. His group and others are working to

better define the effects of halogen bonding in biology and to update potential energy functions for modeling.

Because halogen bonds have a strong geometric requirement, "it's much more difficult in the protein to establish a strong halogen bond than a strong hydrogen bond," says Frances Diederich, a professor of chemistry at the Swiss Federal Institute of Technology, Zurich. "But if you find the right halogen bond and establish it, you have made a big gain in affinity and also selectivity."

Researchers are also looking to explore halogen-like bonds made by elements in groups 16, 15, and 14 of the periodic table. "Covalently bonded atoms in these groups can have two, three, and four σ -holes, respectively, or more if hypervalent," says Peter A. Politzer, an emeritus chemistry professor at the University of New Orleans.

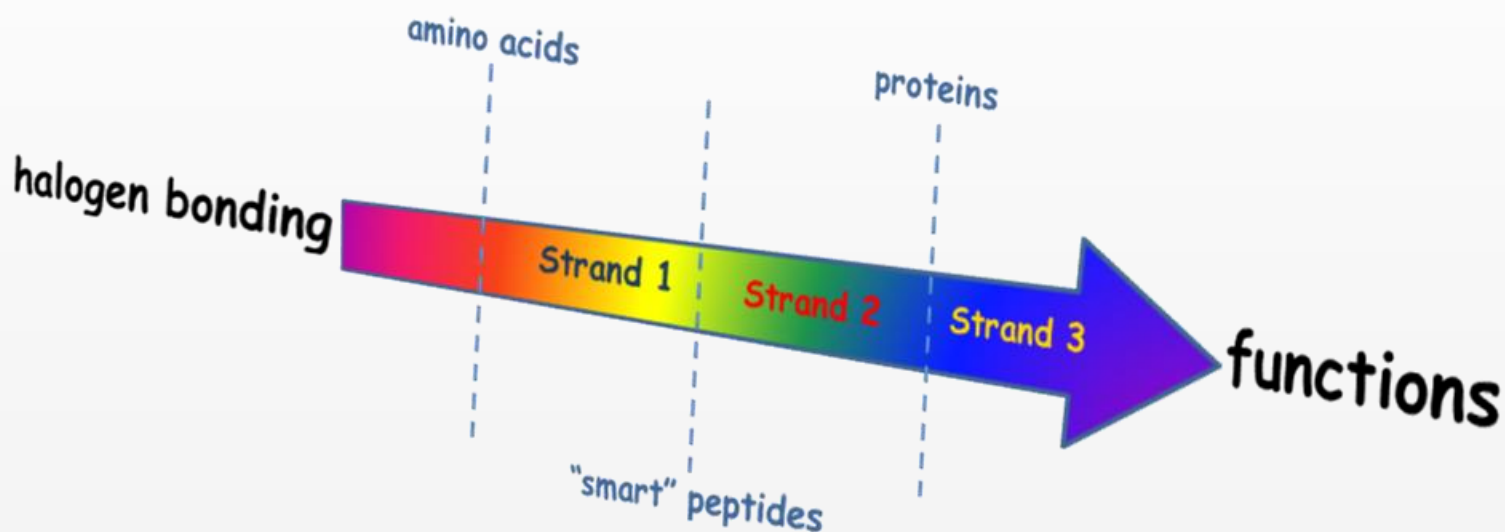
IUPAC is accepting comments on the halogen bond proposal through the end of November. —JULIAN REMBELEY

"If you find the right halogen bond and establish it, you have made a big gain in affinity and also selectivity."

36

A halogen bond occurs when there is evidence of a net attractive interaction between an electrophilic region associated with a halogen atom in a molecular entity and a nucleophilic region in another, or the same, molecular entity.

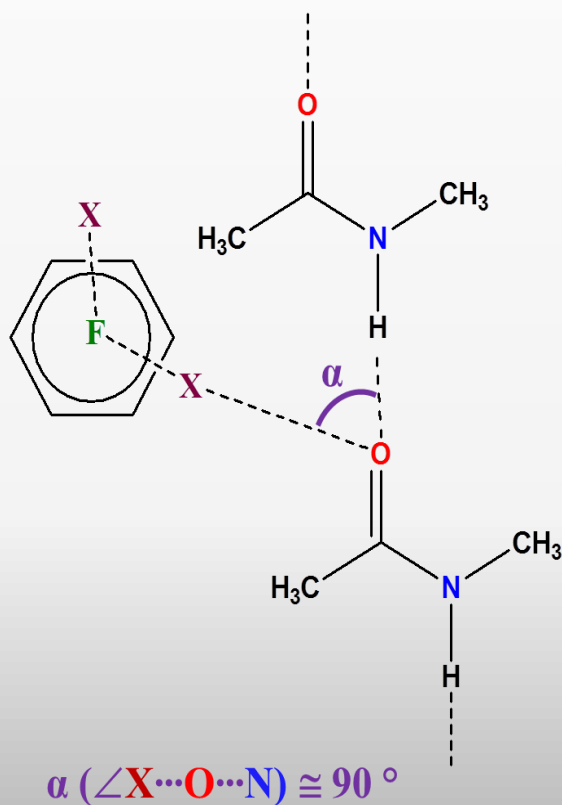
ERC-StG-2012 (Consolidator), PE5-Supramolecular Chemistry



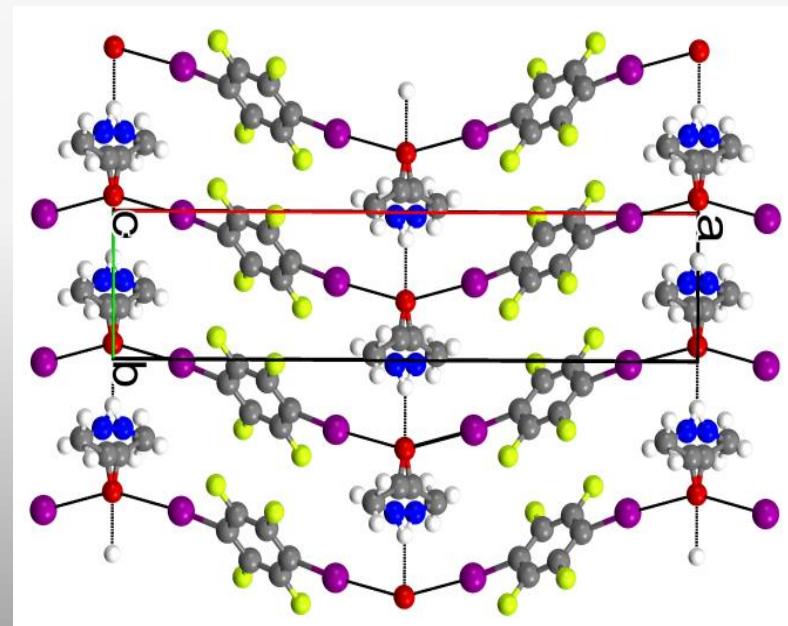
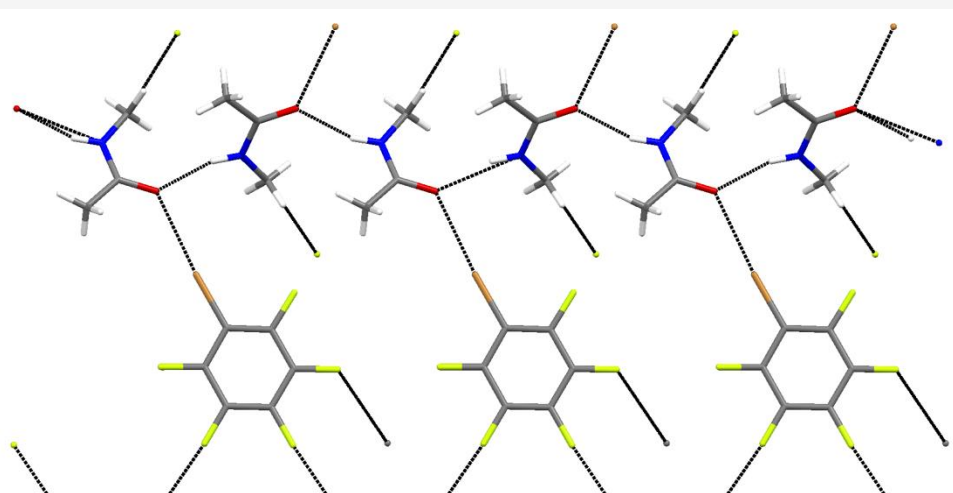
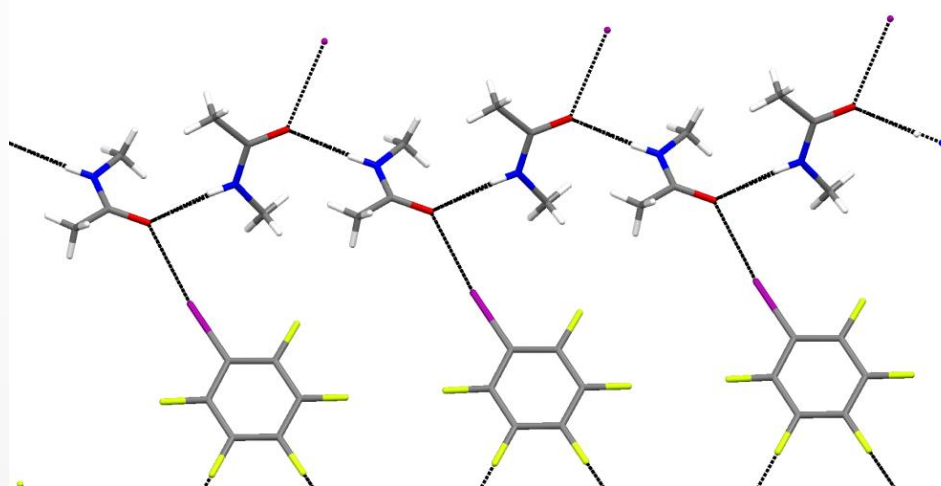
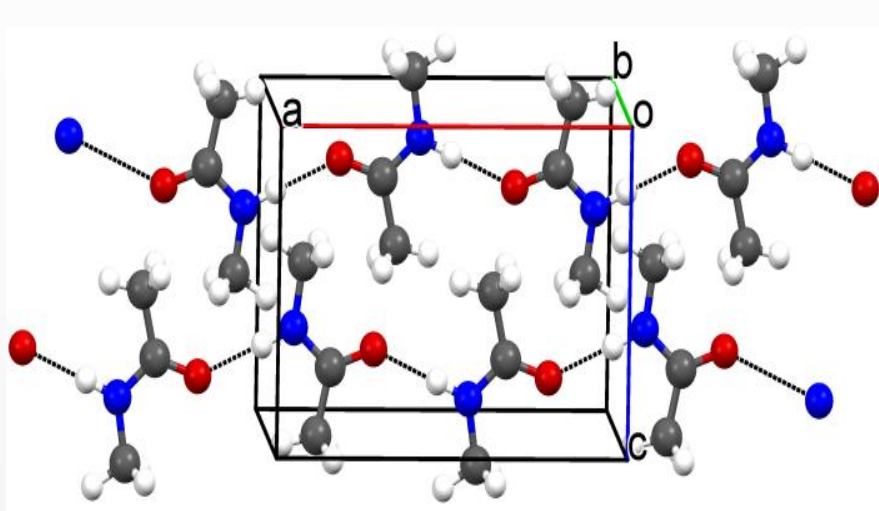
The overall goal of the *FOLDHALO* project will be to fully elucidate the capabilities and properties of halogen atoms as recognition “sticky” sites in the context of biomolecules.

(Project web-site: www.foldalo.it)

Remarkable feature, common for **3a-3f**:
nearly orthogonal angle $\angle \text{X} \cdots \text{O} \cdots \text{N}$

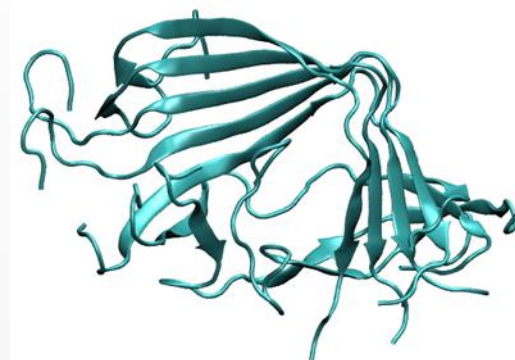
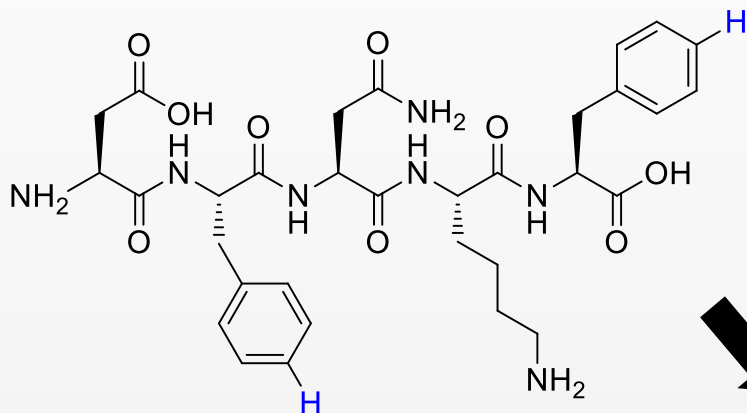


Adduct	$\alpha (\angle \text{X} \cdots \text{O} \cdots \text{N}), ^\circ$
3a	97.2
3b	91.9
3c	81.2
3d	82.8; 88.9 ^a
3e	81.9
3f	78.9
^a two symmetry independent molecules	



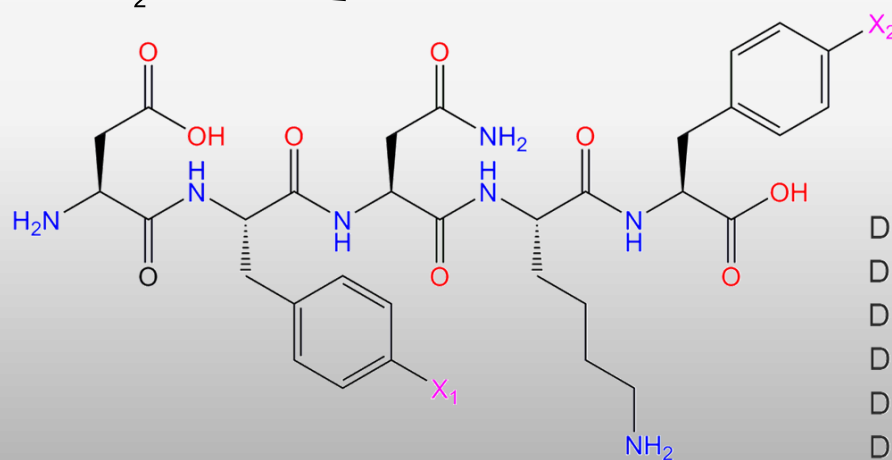
Fibrillation of human calcitonin plays a role in thyroid medullar carcinoma

H₂N-CGNLSTCMLGTYTQ**DFNKF**HTFPQTAIGVGAP-COOH

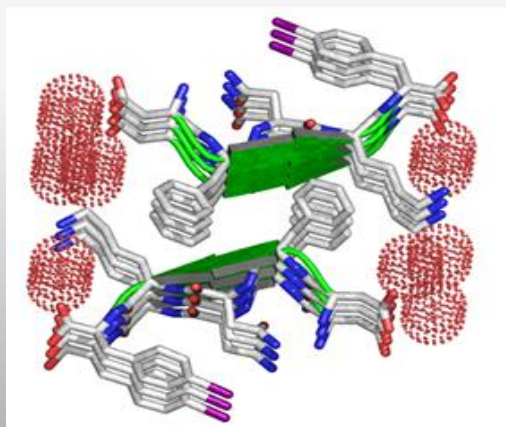


J. Biol. Chem. 2017, 292, 7348.

halogenation



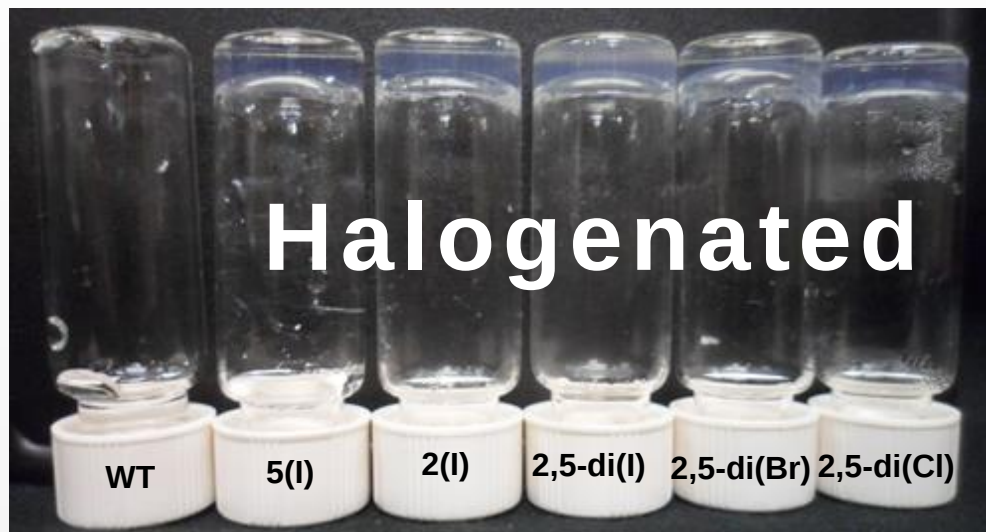
		X ₁	X ₂
DFNKF	1	H	H
DFNKF(I)	2	H	I
DF(I)NKF	3	I	H
DF(I)NKF(I)	4	I	I
DF(Br)NKF(Br)	5	Br	Br
DF(Cl)NKF(Cl)	6	Cl	Cl



Chem. Eur. J. 2017, 23, 2051.
(VIP Paper and Front Cover).

hCT₁₅₋₁₉ NH₂-Asp-Phe-Asn-Lys-Phe-COOH
1 2 3 4 5

Finding the best candidates



Peptide	Transition T (°C)
5(I)	64.7-69.0
2(I)	88.2-92.7
2,5-di(I)	114.0-119.0
2,5-di(Br)	92.1-96.4
2,5-di(Cl)	78.1-82.4



Peptide	Peptide Sequence	Minimum Critical Concentration (mM)		Time required for gelation
1	WT	75	5%	96h
2	5(I)	7.5	0.6%	96h
3	2(I)	2.5	0.2%	168h
4	2,5-di(I)	5*	0.5%	1h
5	2,5-di(Br)	8	0.7%	3h
6	2,5-di(Cl)	5	0.4%	360h

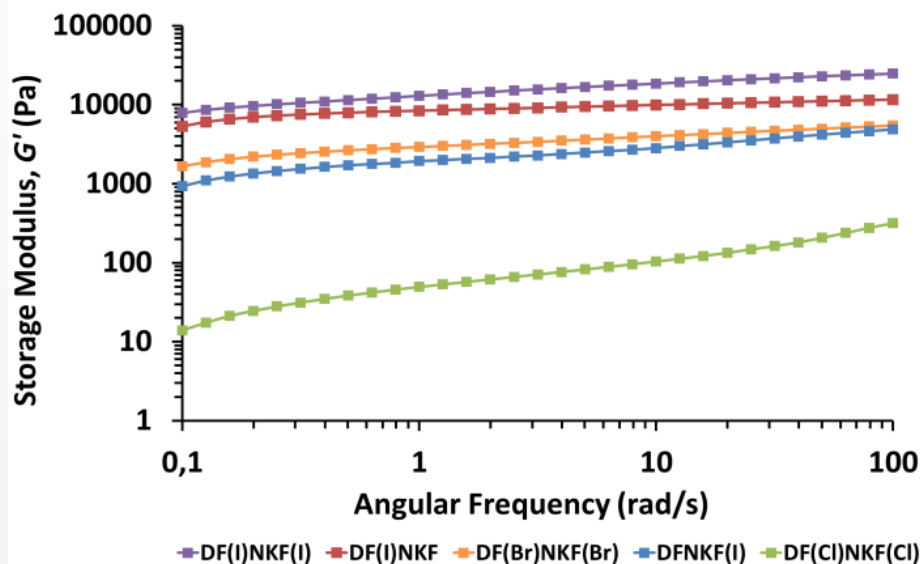
$2,5\text{-di(I)} > 2,5\text{-di(Br)} \geq 2\text{(I)} > 2,5\text{-di(Cl)} \geq 5\text{(I)} \gg \text{WT}$



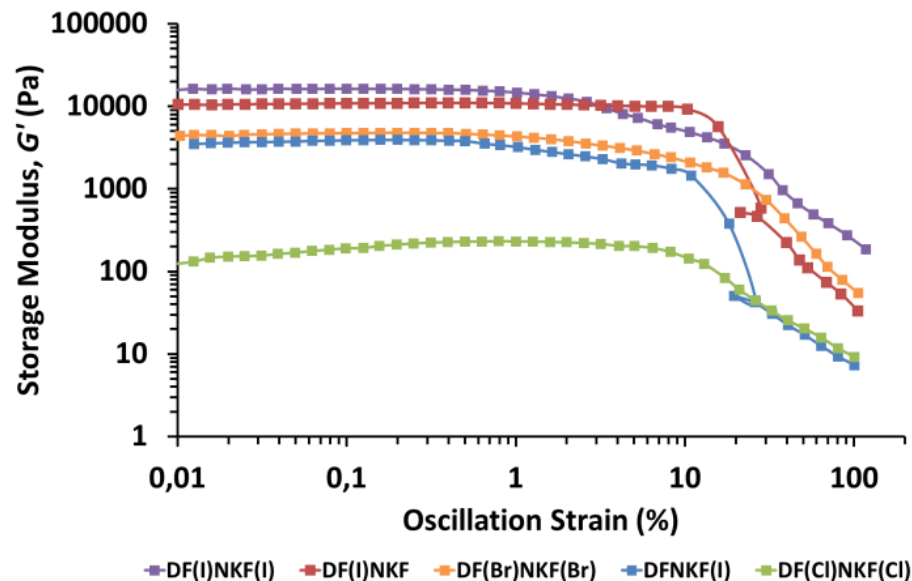
Oscillatory Rheology by Using the Ring-cast Method



Frequency Sweep



Amplitude Sweep



$G' > 10^4$ Pa

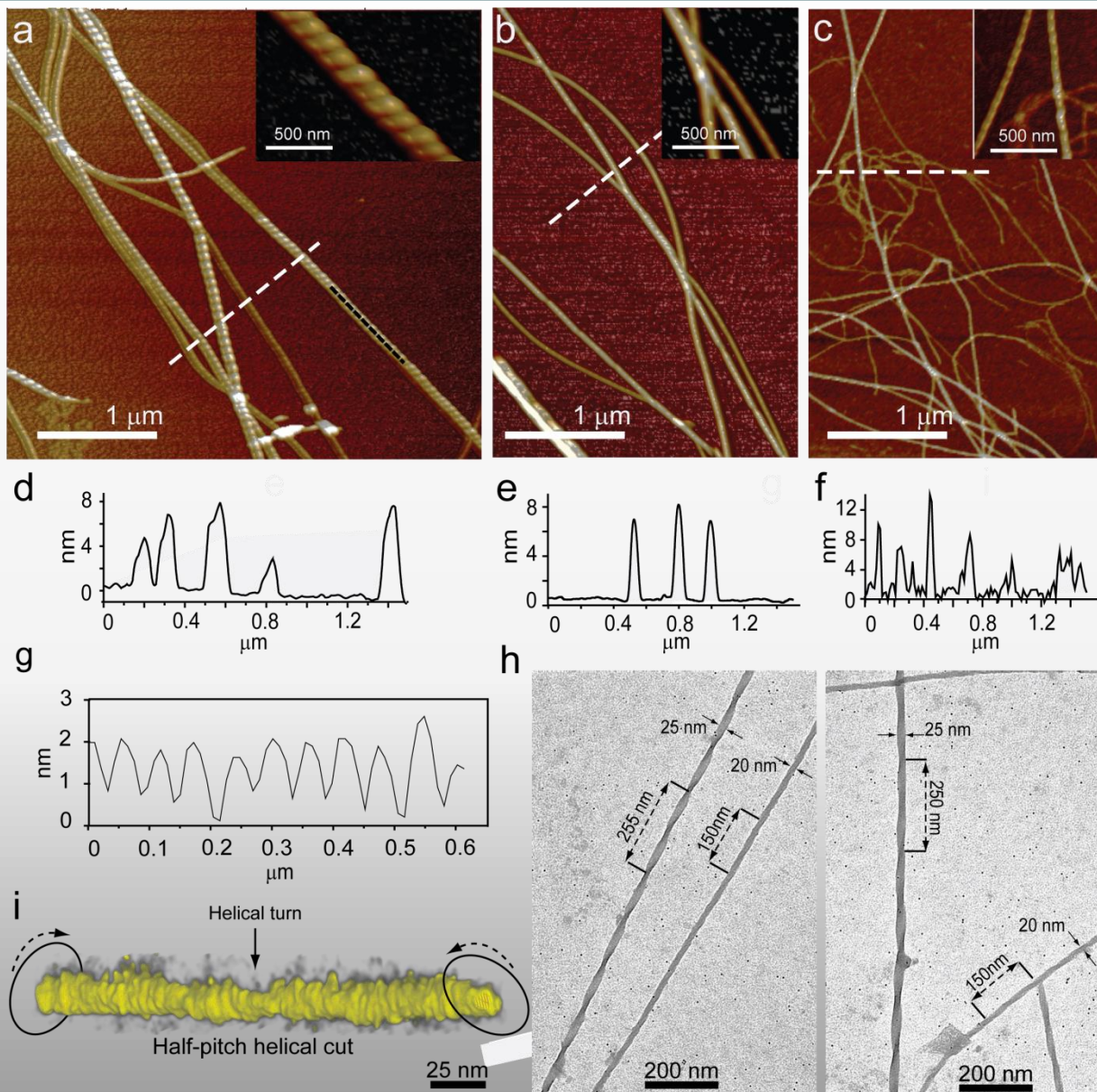
2,5-di(I) > 2(I) > 2,5-di(Br) > 5(I) > 2,5-di(Cl)

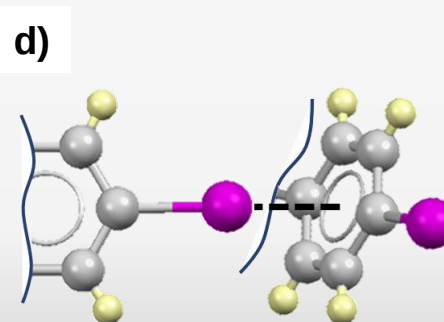
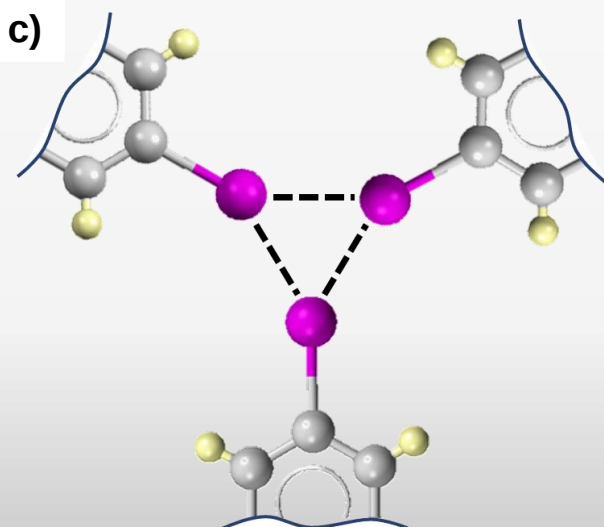
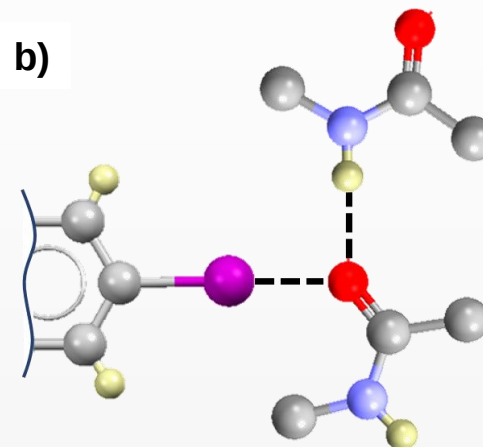
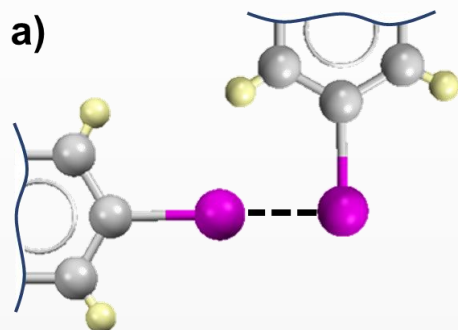


Nature Comm. 2015, 6:7574, DOI: 10.1038/ncomms8574.

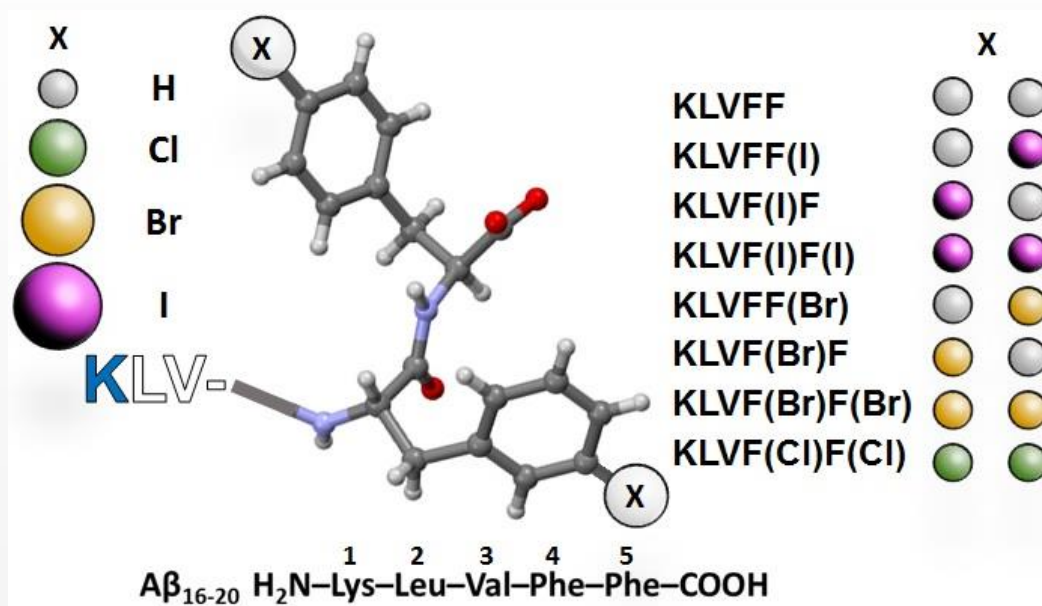


Halogenated Fibers Self-Assemble into Helical Twisted Ribbon-like Superstructures

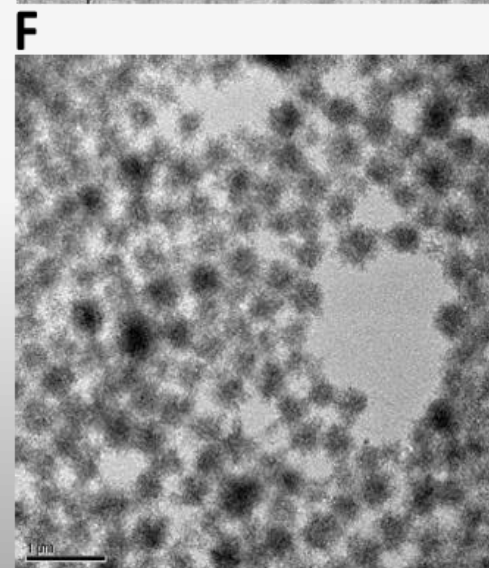
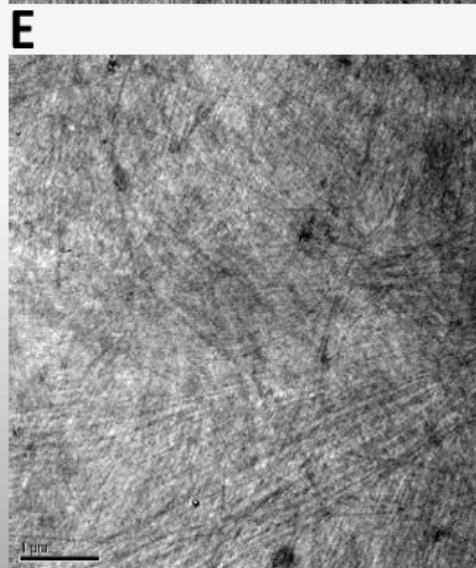
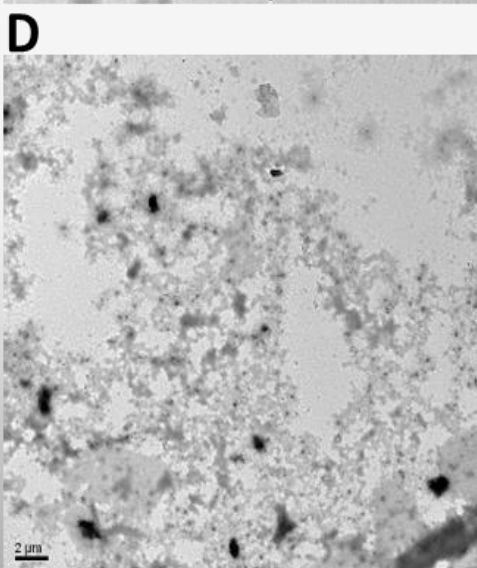
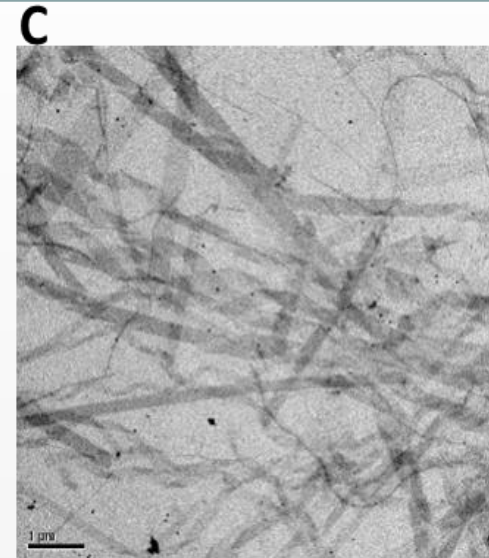
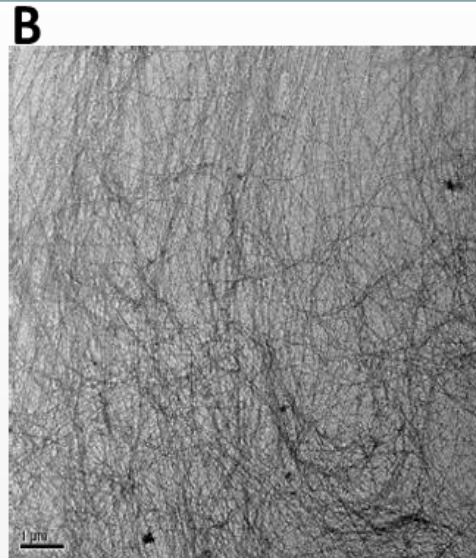
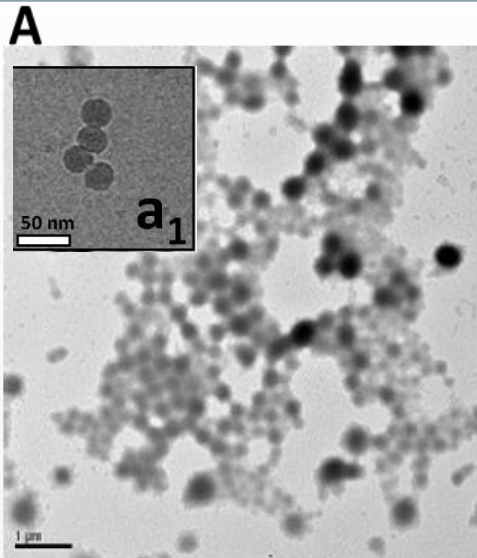


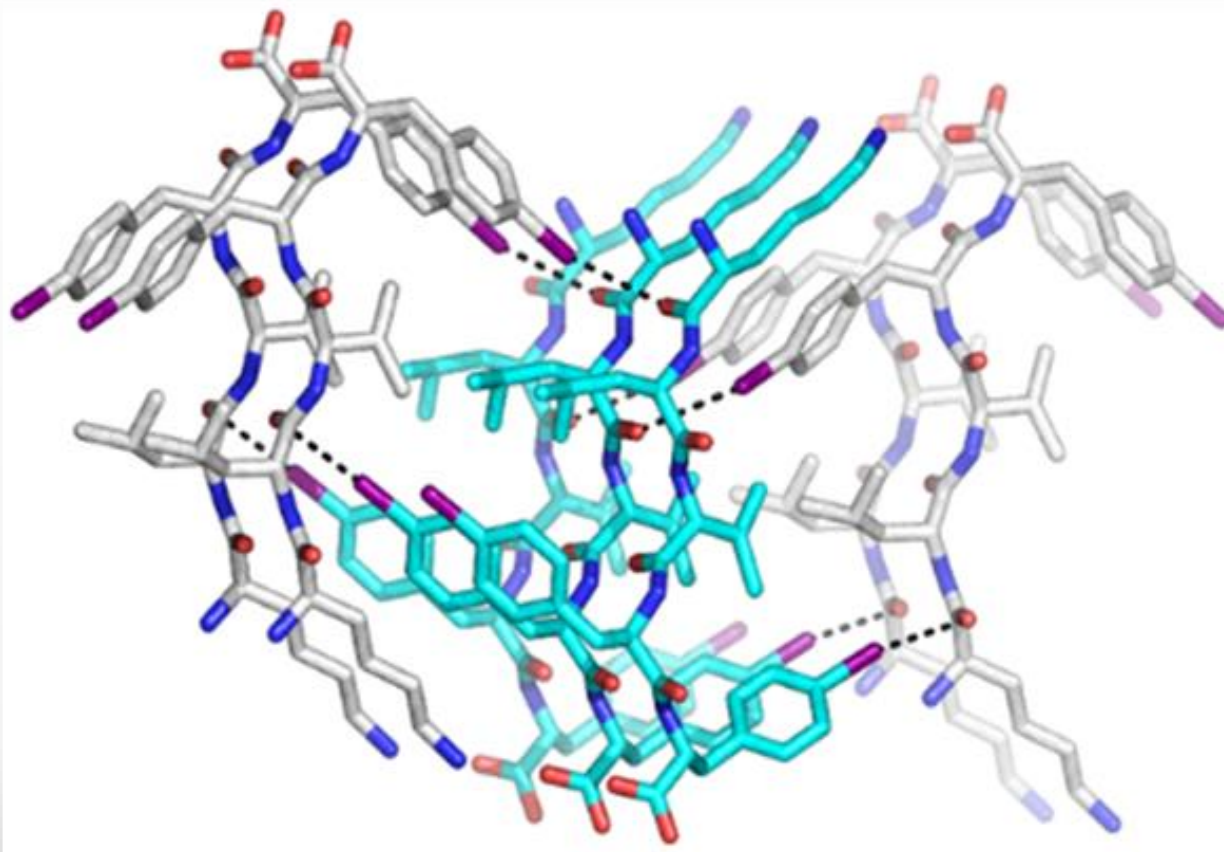


Interactions involving halogen atoms of peptide **2-6** as hydrogen bond acceptors may be ruled out since Cl and Br should give stronger interactions, and fibril formation effects, than I. Interactions involving the ionisable side chains of the peptides, Asp and Lys, may also be ruled out since the formation of these hydrogels is largely pH-independent. Finally, it is unlikely that aromatic $\pi \cdots \pi$ interactions are playing an important role in the hydrogel self-assembly here, since this would not explain the observed dependence on halogen polarizability. Halogen bonding therefore results the most likely interaction responsible for the observed fibril formation effects.



(15 mM Water Solutions)





$O_1 \cdots I_5$ 3.34(3) Å and $O_2 \cdots I_4$ 3.43(3) Å, with $C=O \cdots I$ and $O \cdots I-C$ angles of $111(2)^\circ$ and $103(2)^\circ$, and $165(1)^\circ$ and $159(1)^\circ$, respectively. The $O_2 \cdots I_4$ halogen bond occurs orthogonally²⁷ to the $O_2 \cdots N_3$ hydrogen bond of the β -sheet ($N \cdots O$ distance 2.99(3) Å; $N \cdots O \cdots I$ angle $88.5(8)^\circ$).

Invertebrates modify the mechanical properties and structures of their biomaterials either via calcification or by the introduction of heavy elements (Br, Zn, Cl, I).

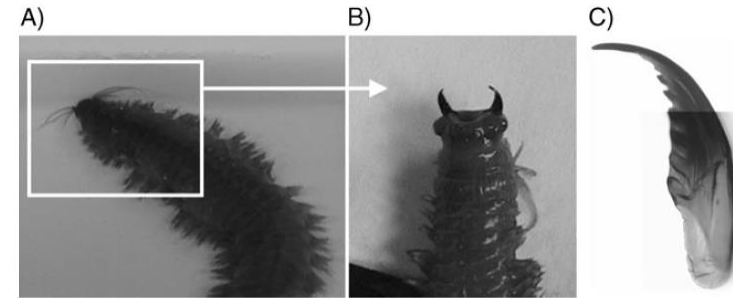
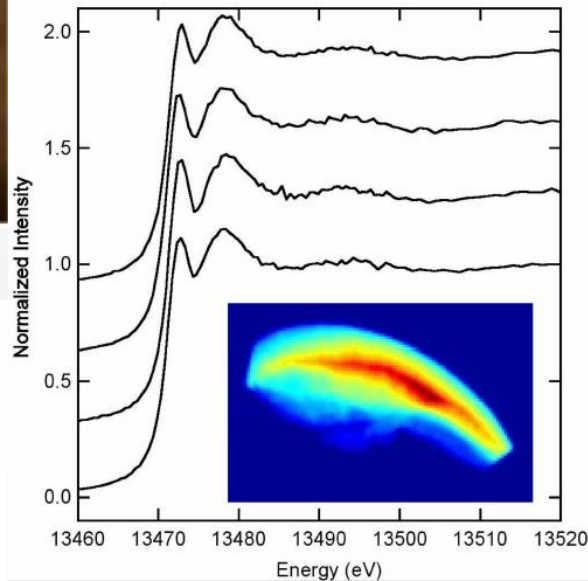
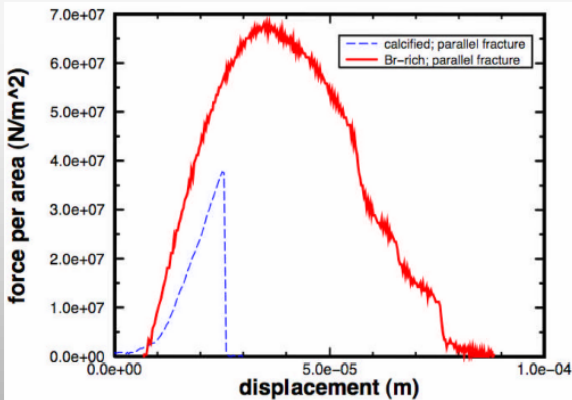
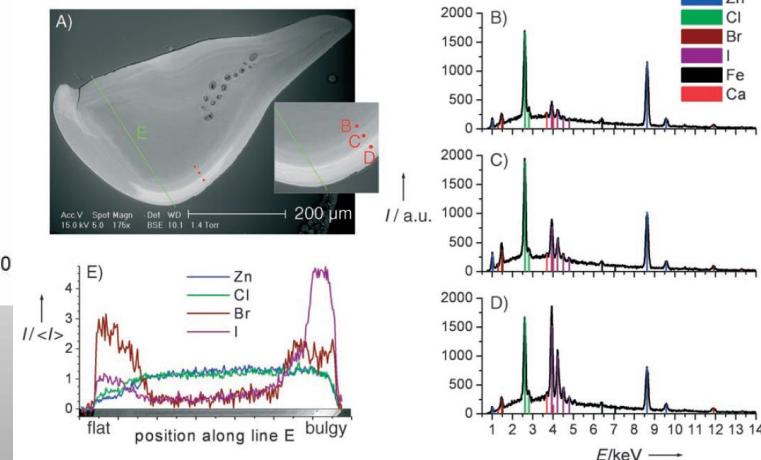
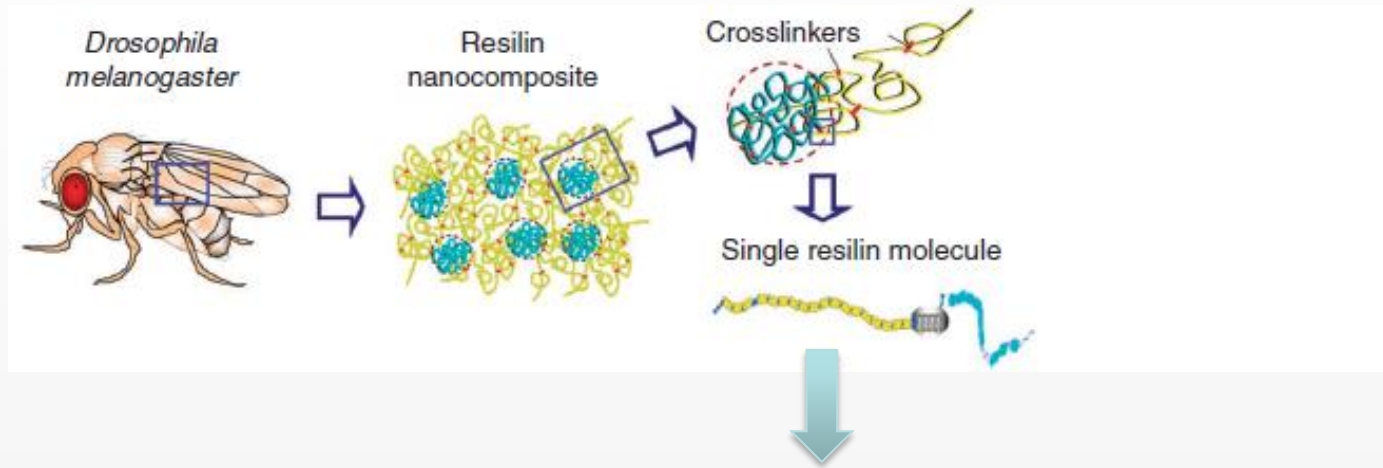


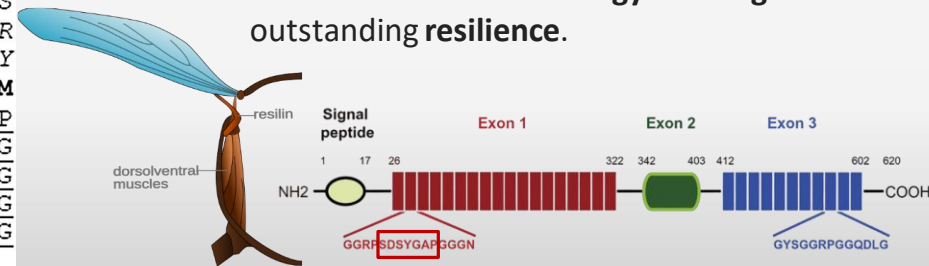
Figure 1. *N. virens* worm A) shown exploring, and B) with everted proboscis; the jaws are shown from the ventral side. C) A jaw extracted from the animal.





1	MVRPEPPVNS	YLPPSDSYGA	PGQSGPGGRP	SDSYGAPGGG	NGGRPSDSYG
51	APGQGQGQGQ	GQGGYAGKPS	DTYGAPGGGN	GNGGRPSSSY	GAPGGGNGGR
101	PSDTYGAPGG	GNGGRPSTDY	GAPGGGGNGN	GGRPSSSYGA	PGQGQNGNG
151	GRSSSSYGAP	GGGNGGRPSD	TYGAPGGGNG	GRPSTDYGA	GGGNNGGRPS
201	SSYGAPGGGN	GGRPSTDYGA	PGGGNGNGSG	GRPSSSYGAP	GQGQGGFGR
251	ESDSYGAPGQ	NQKPSDSYGA	PGSGNGNGGR	PSSSYGAPGS	GPGGRPSDSY
301	GPPASGSGAG	GAGGSGPAGA	DYDNDEPAKY	EFNYQVEDAP	SGLSFGHSEM
351	RDGDFTTGQY	NVLLPDGRKQ	IVEYEADQOG	YRPQIRYEGD	ANDGSGPSGP
401	GGPGGQNLGA	DGYSSGRPGN	GNGNGNGGYS	GGRPGGQDLG	PSGYSGGRPG
451	GQDLGAGGYS	NGKPGGQDLG	PGGYSGGRPG	GQDLGRDGYS	GGRPGGQDLG
501	ASGYSNRPG	GNGNGGSDGG	RVIIGGRVIG	GQDGGDQGY	GGRPGGQDLG
551	RDGYSSGRPG	GRPGGNGQDS	QDGGQGYSSGR	PGQGGRNGFG	PGGQNGDNDG
601	SGYRY				

Resilin is a **native elastomeric protein** which features efficient **energy storage** and outstanding **resilience**.



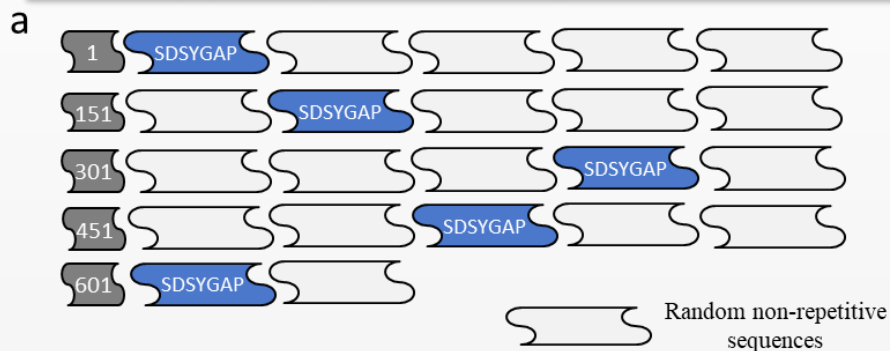
Resilin sequence ~ 600 amino acids



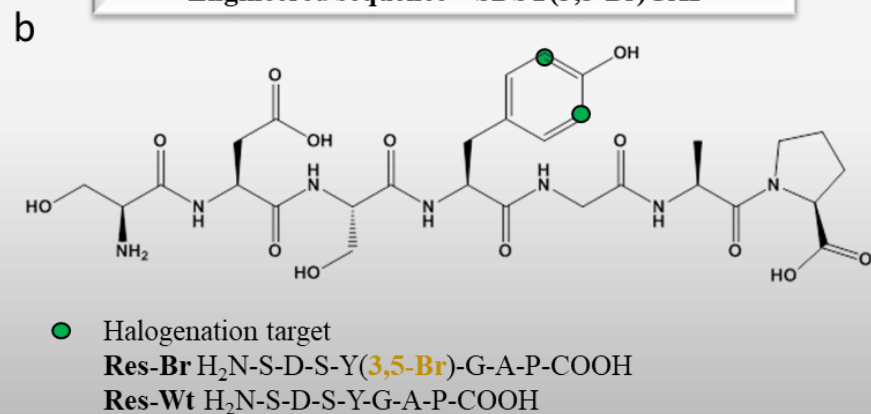
Halogenation of Elastomeric Peptides



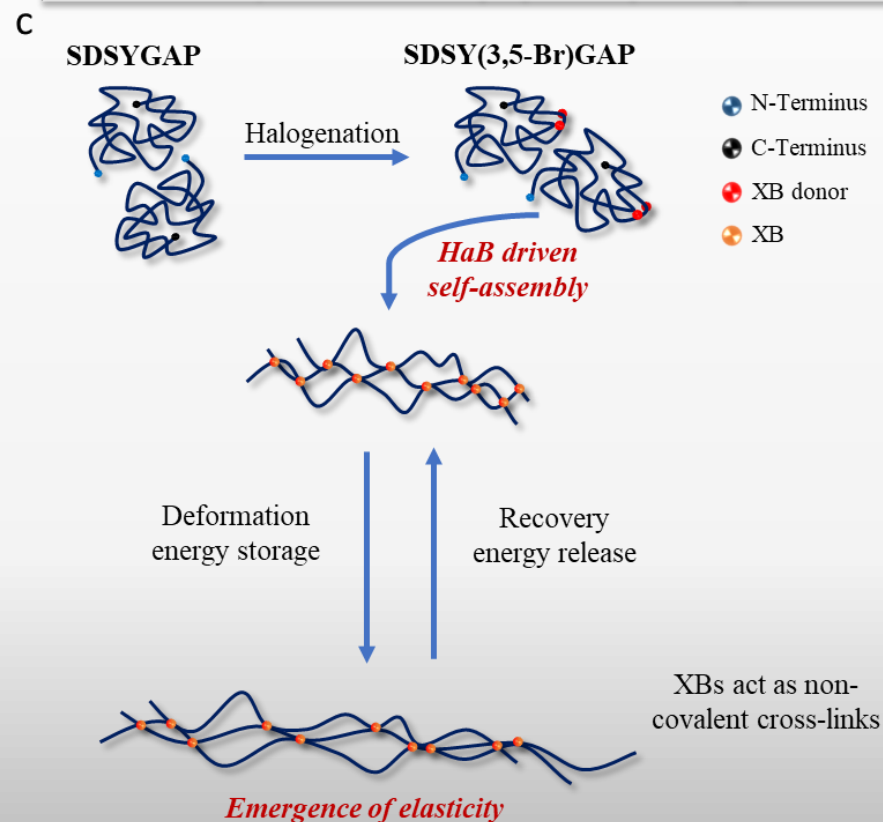
Small peptide fragment from full-length resilin - SDSYGAP



Engineered sequence – SDSY(3,5-Br)GAP



Self-assembly and elastomeric properties upon halogenation

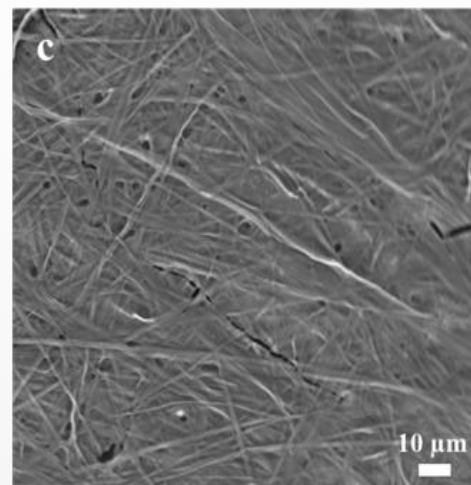
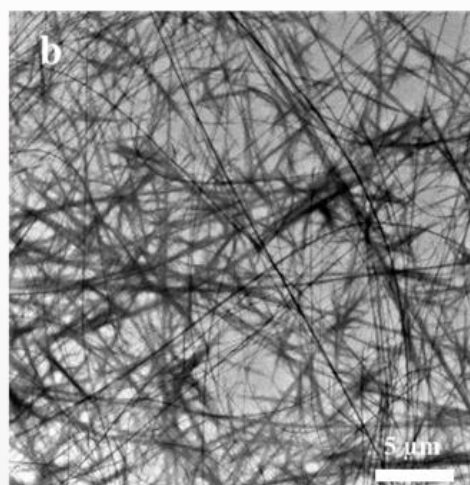




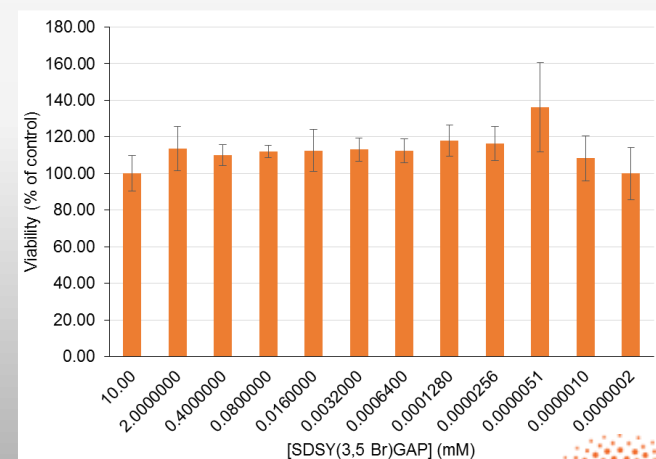
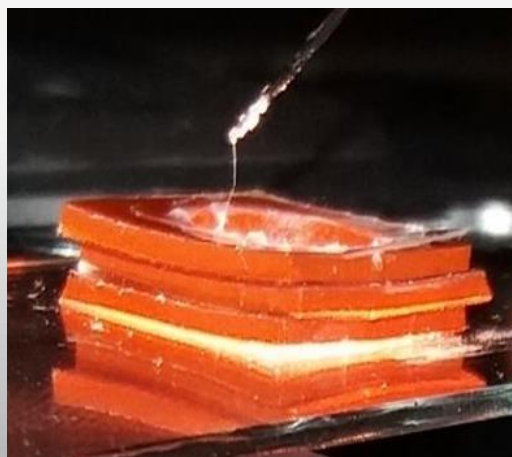
Halogenation of Elastomeric Peptides



MINIRES



[WO2019116342A1](#)

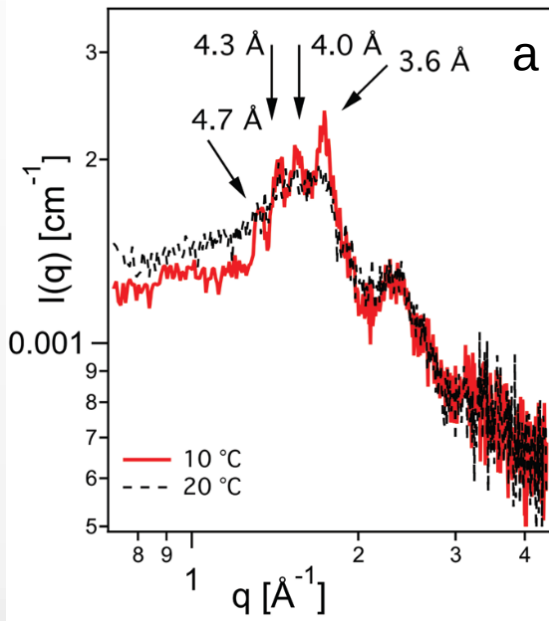


In collaboration with Markus Linder, Aalto University, Helsinki, Finland.

PoC2017 “A Minimalist Peptide Elastomer” – Switch2Product Innovation Challenge 2017 Champion

200 mM **Res-Br** hydrogels and **Res-Wt** solutions

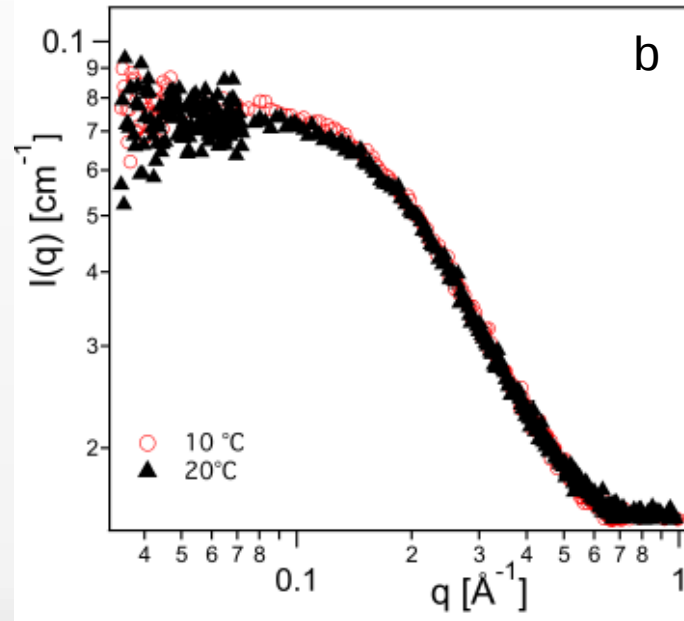
WAXS



↑

q values among $1\text{--}4\text{ Å}^{-1}$
Four distinct peaks correlated with a structural order typical of an amyloid cross- β architecture

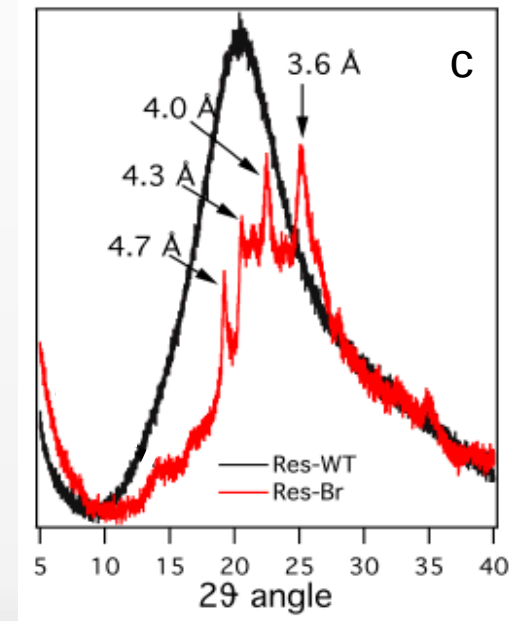
SAXS



↑

The scattering profile of the wild-type sequence is typical of unstructured proteins

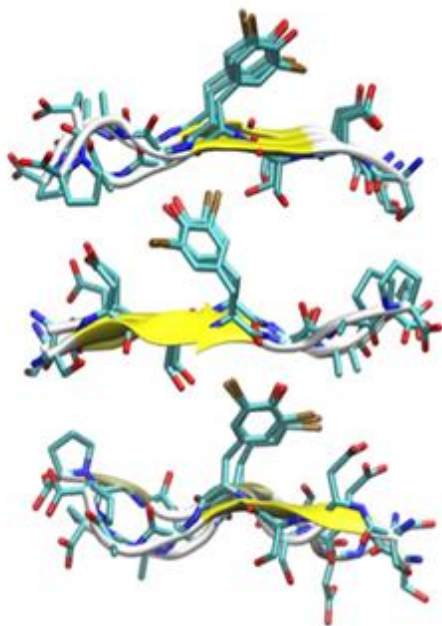
PXRD



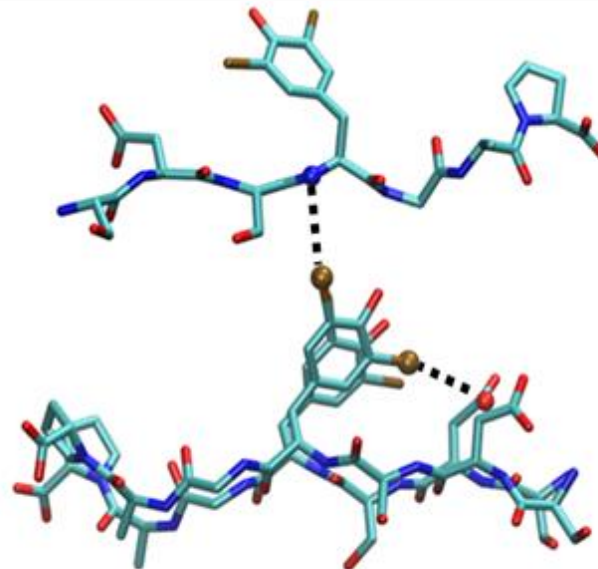
↑

Shows **diffraction peaks** consistent with a **cross- β architecture typical of amyloid materials**

a

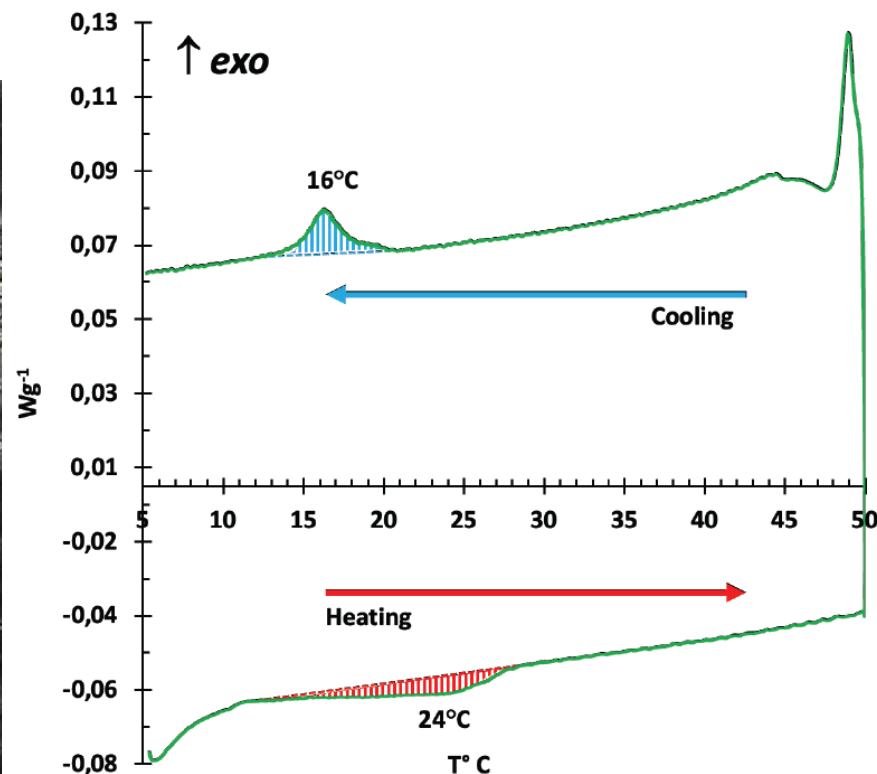
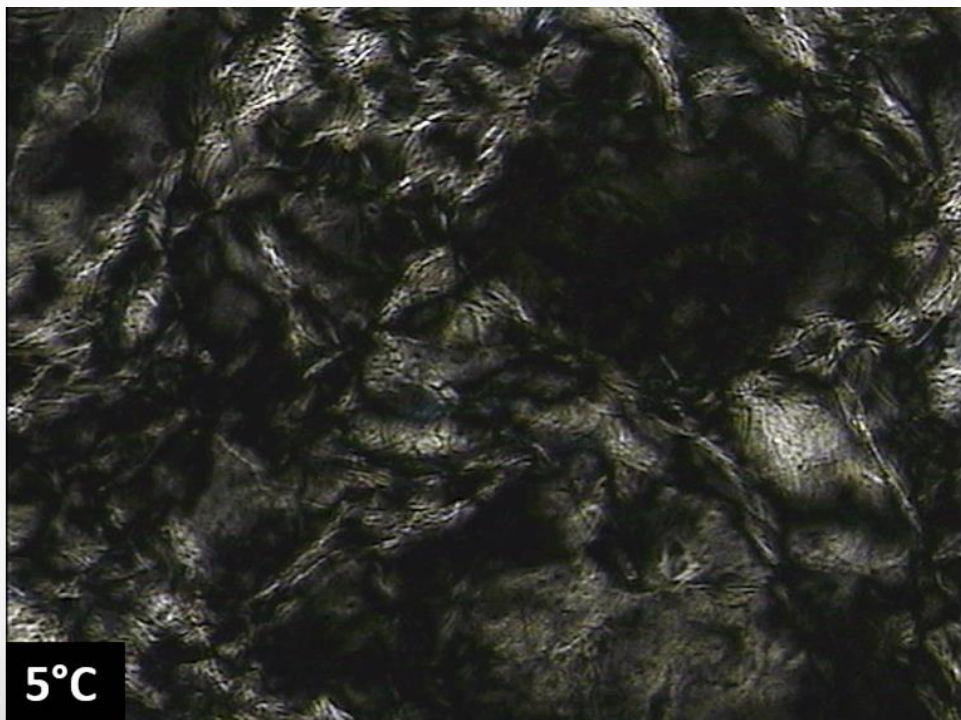


b



- In the PAN (Parallel, Antiparallel, C-to-N) configuration, **parallel β -sheets** of Res-Br are stabilized by hydrogen bonds, in addition to **intermolecular $\pi \cdots \pi$ stacking between paired di-bromo-Tyr rings**
- Bromine atoms exert a key role in strengthening peptide self-assembly** thanks to XBs.

Heating from 5°C to 32°C



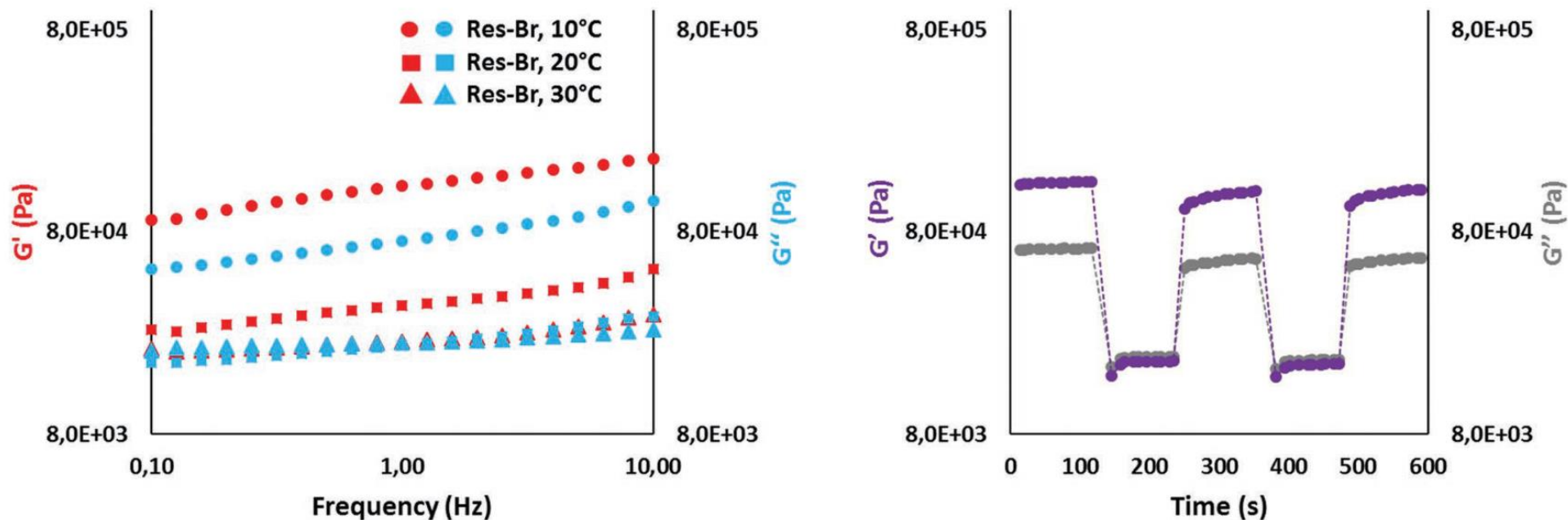
Polarized optical microscopy (POM):

The birefringent self-assembled peptide network progressively dissolves with increasing temperature

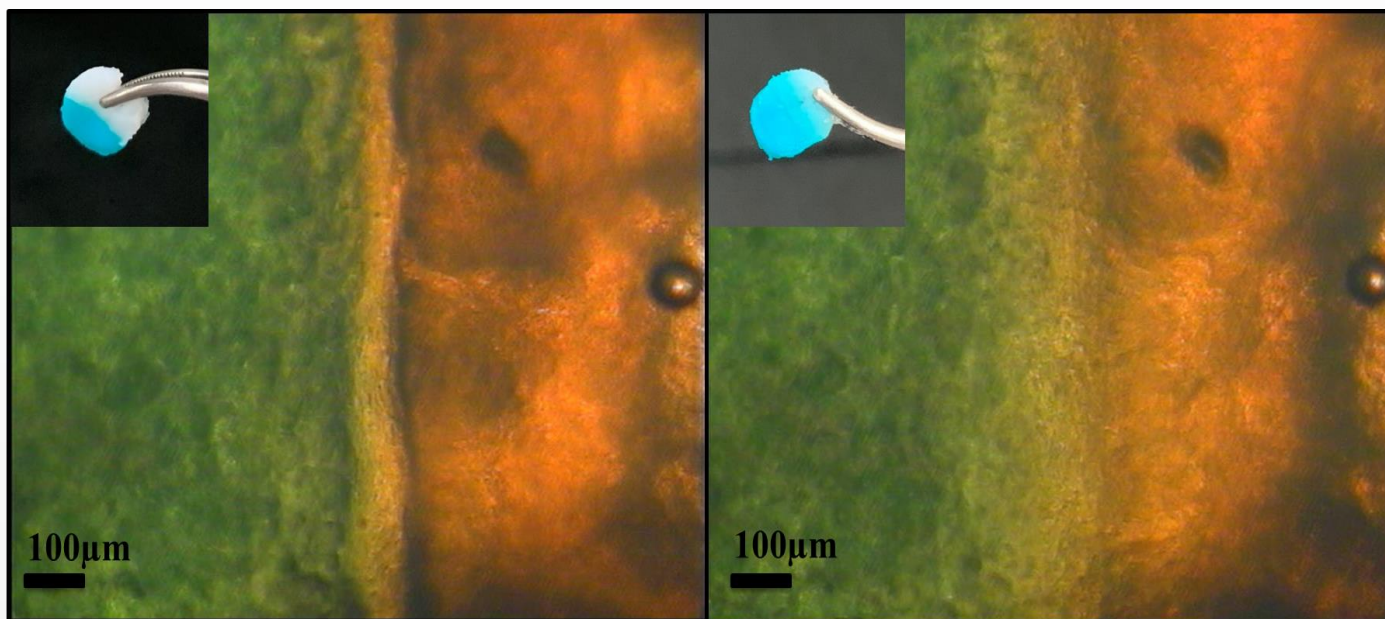
Differential Scanning Calorimetry (DSC):

Endothermic transition close to room temperature (24°C), consistent with **peptide solubilization** hence **disassembly of the supramolecular architecture**. Formation of the gel network is **thermoreversible**, with an exothermic signal in the range 18-15 °C

200 mM gel of brominated peptide; analyzed at controlled temperature



- **Predominance of elastic component ($G' > G''$)** for the tested hydrogels
- The **elastic modulus decreases with temperature**
- Step test to describe the regeneration of the material after deformation; **recovery interval there is a rapid and almost total recovery of the structure**

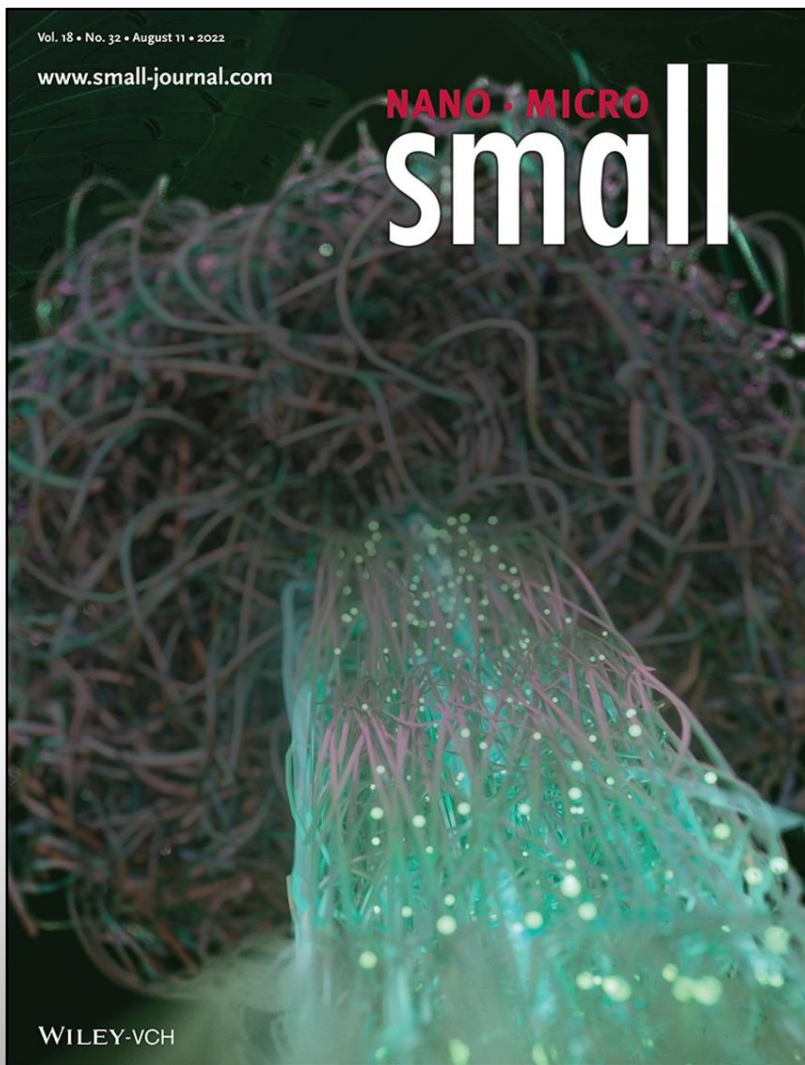


The microscopy images show the initial (left) and final (right) stage of the healing process, with the **disappearance of the joint between the two pieces of hydrogel**. The overall process takes **5 minutes**.

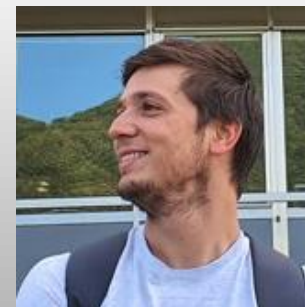
The insets show the macroscopic appearance of the **joint pieces of gel at the initial stage (left) and 24 h later (right)**. After healing, diffusion of the blue dye occurs in the whole sample.



A. Pizzi



L. Sori



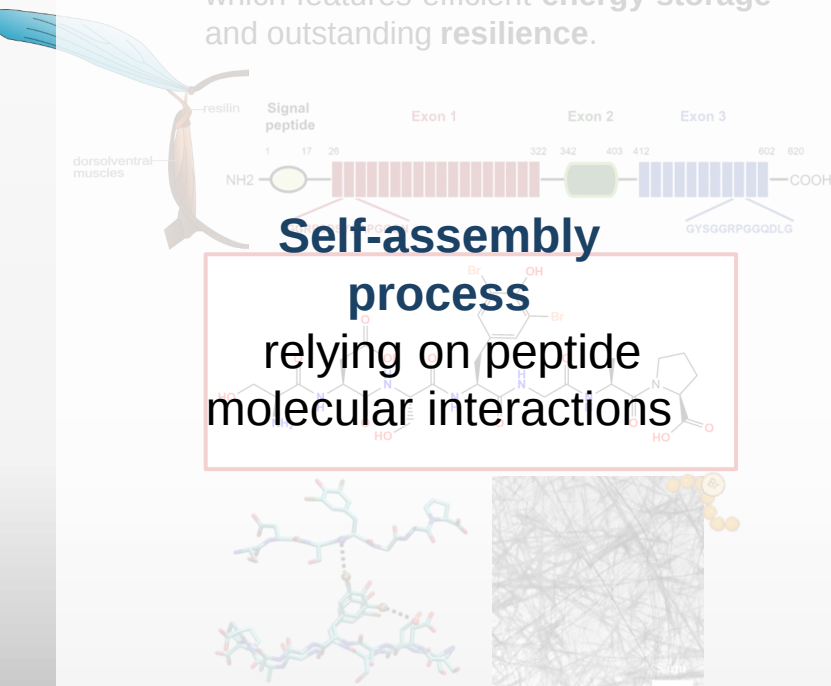
A. Pizzi, L. Sori, *et. al.*,
***Small* 2022, 18, 2200807.**

A resilin-derived heptapeptide



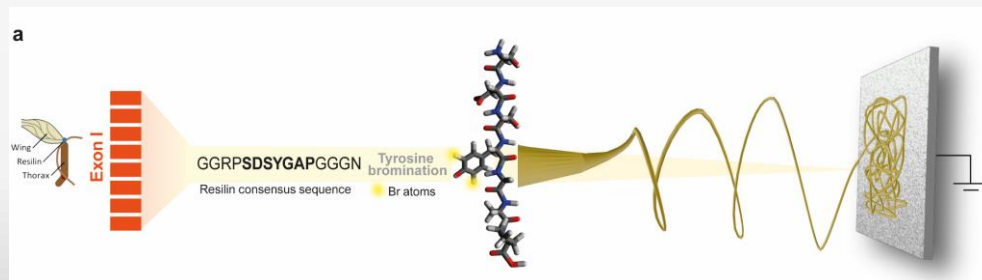
E. Marelli

Resilin is a **native elastomeric protein** which features efficient **energy storage** and outstanding **resilience**.



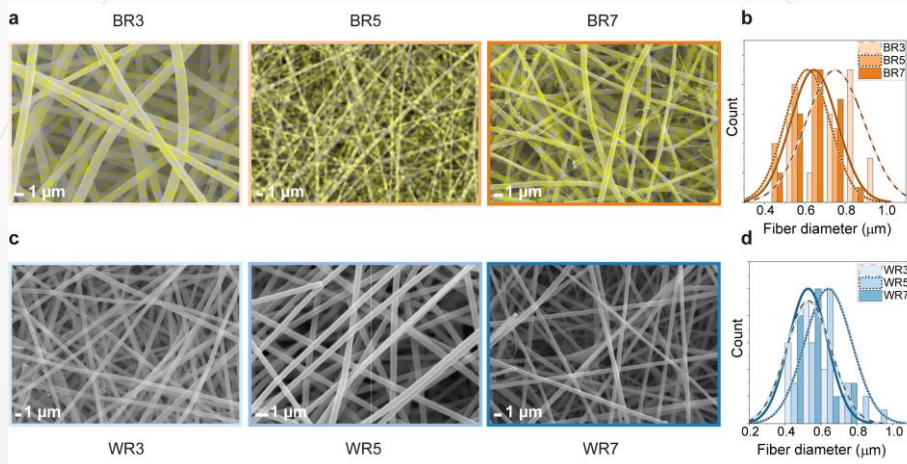
Small, 2022, 18, 2200807

Directed assembly can be imposed through **external constraints** forming macroscopic materials

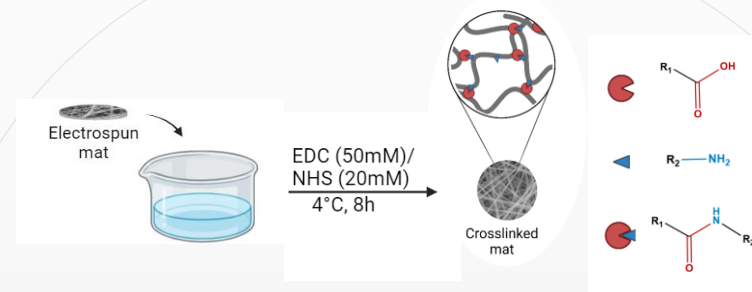


Use of **electrospinning** to direct the assembly of SDSY(3,5Br)GAP into nanofibers with higher control over material microstructure.

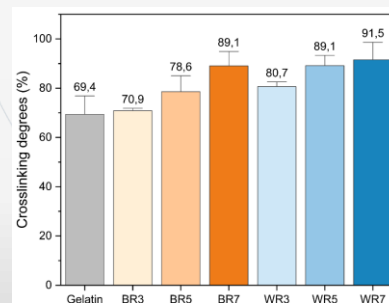
Electrospinning and crosslinking procedures



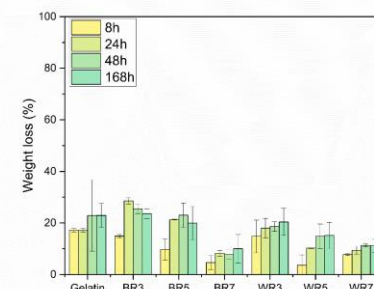
- Concentration 15% *m/v*-30% *m/v*
- Solvent:
 $\text{CH}_3\text{COOH}:\text{H}_2\text{O}$ 1:1
- Voltage: 12-15kV
- Flow rate: 254 $\mu\text{L/h}$
- Distance from collector: 12 cm



Crosslinking degrees evaluated via TNBS

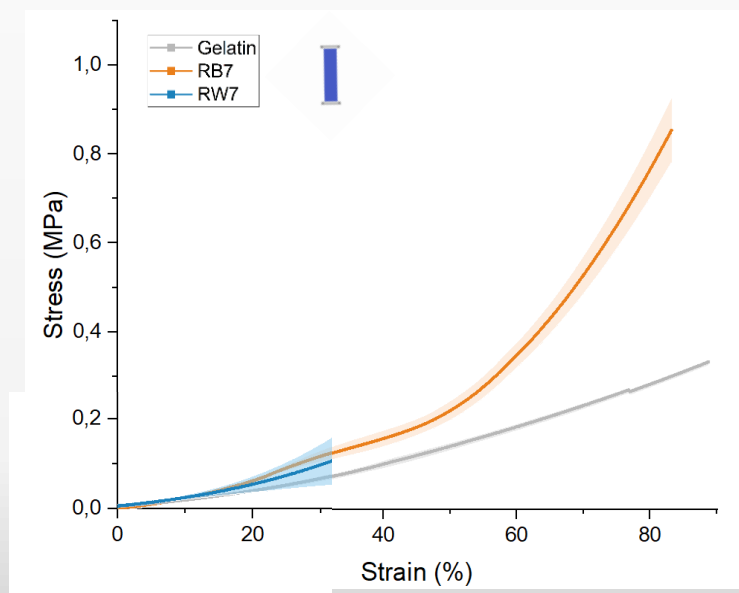


Scaffold stability evaluated in PBS pH 7.4 up to 7 days



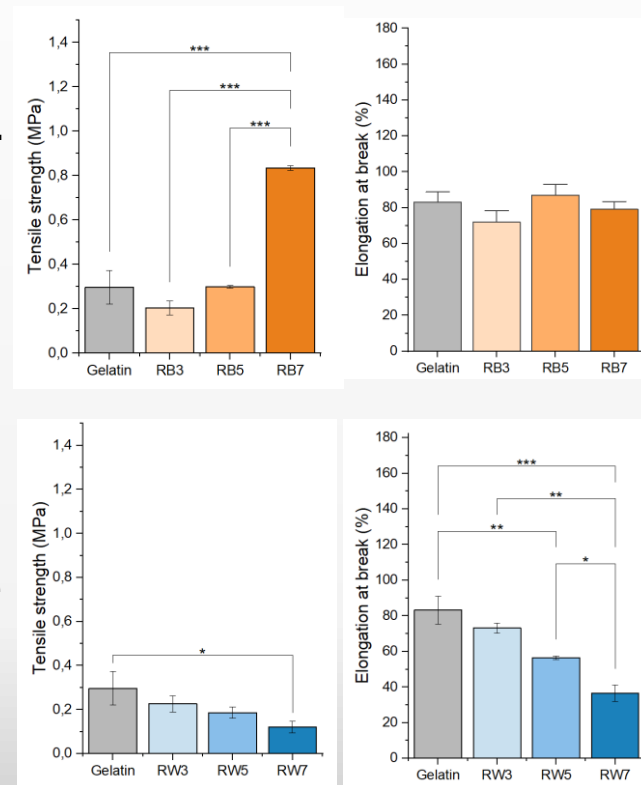
Static mechanical properties

The mechanical properties of the matrices were tested **via tensile testing** after hydration in milliQ for one hour.

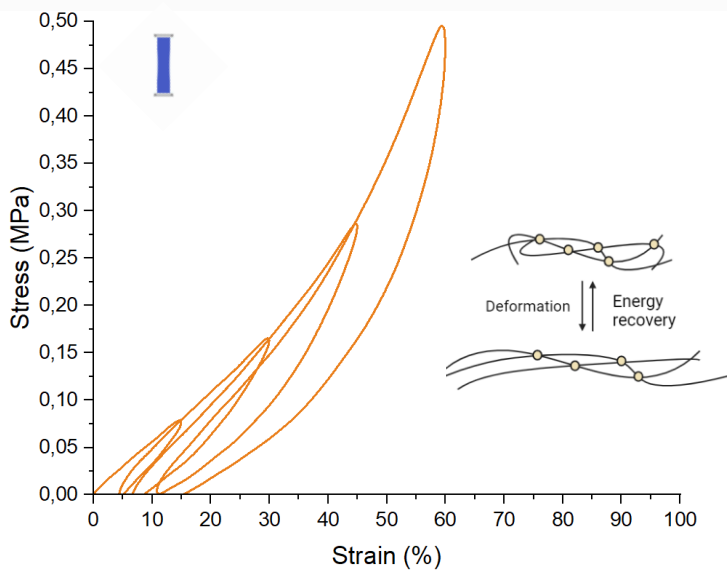


- BR7 **strain-hardening behaviour**

- WR7 **brittle** and prone to fracture



Step-cycle mechanical properties



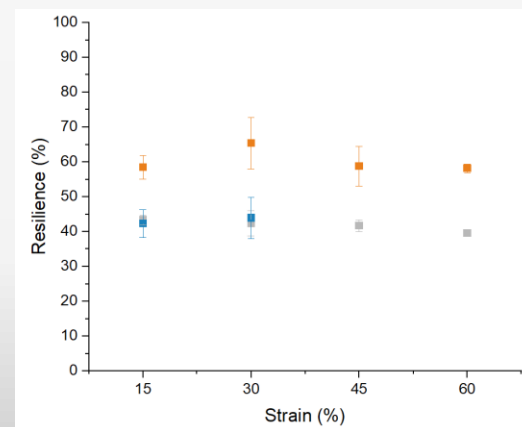
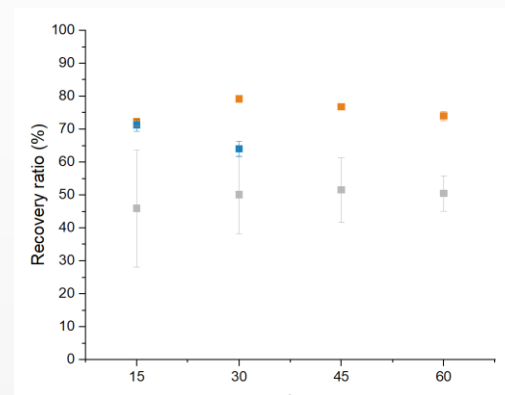
- Good recovery ratio
- Relatively low hysteresis
- Resilience > 60%

Recovery ratio

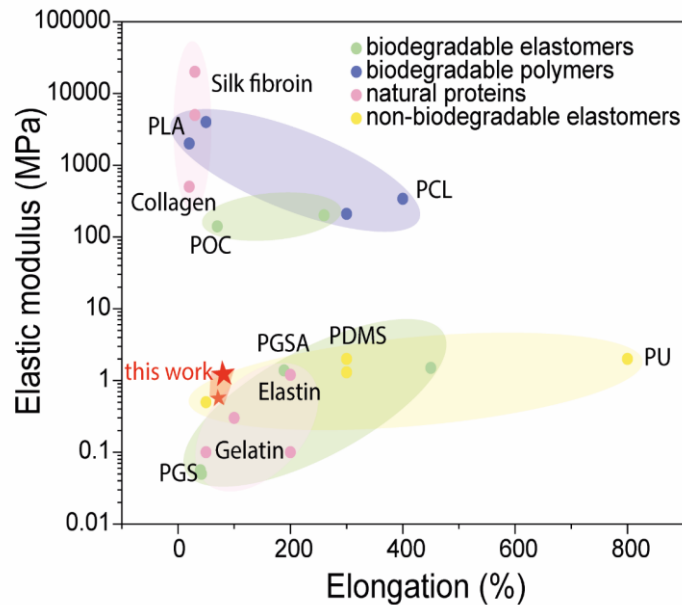
$$\frac{\epsilon_{\max} - \epsilon_{\text{fin}}}{\epsilon_{\max}}$$

Resilience

$$\frac{\text{Area}_{\text{unloading curve}}}{\text{Area}_{\text{loading curve}}}$$



Step-cycle mechanical properties



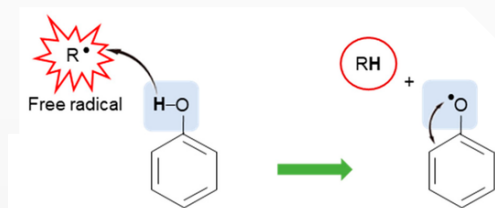
BR7 scaffolds as a promising alternative to conventional bioelastomers



Functional properties

Antioxidant properties

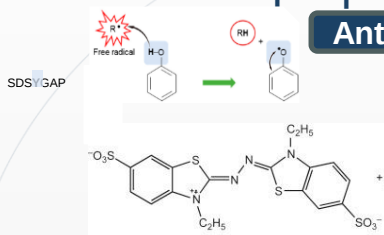
SDS YGAP



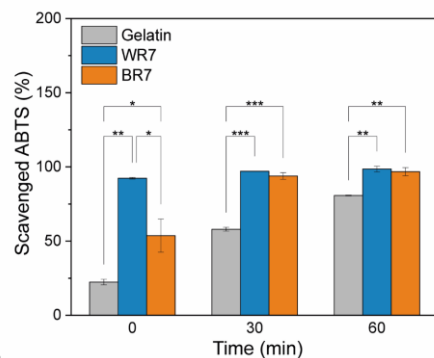
Tyrosine residues are usually associated with antioxidant activity, primarily due to their phenolic structure

Functional properties

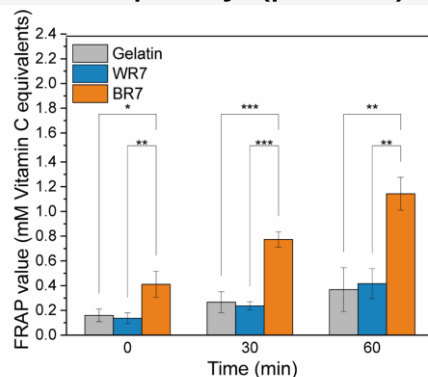
Antioxidant properties



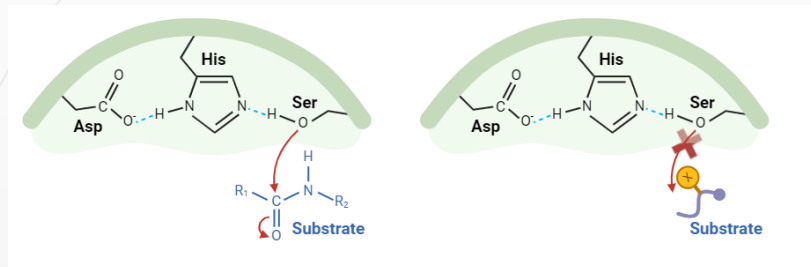
Inhibition of ABTS radicals (pH 7.4)



Fe³⁺ reducing capacity (pH 3.6)



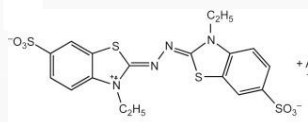
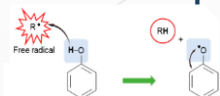
Proteolytic stability



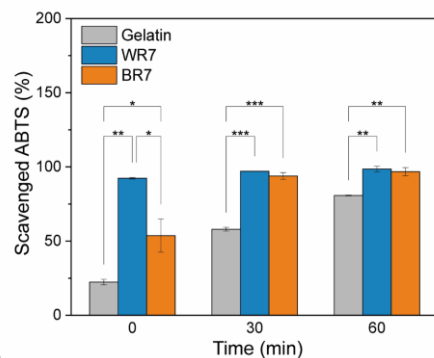
Substituting **natural amino acids** with **halogenated derivatives** can alter interactions with enzymes, potentially **enhancing their stability** against protease degradation.

Functional properties

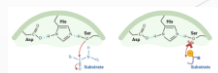
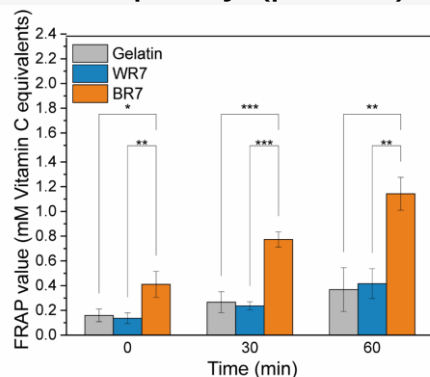
Antioxidant properties



Inhibition of ABTS radicals (pH 7.4)

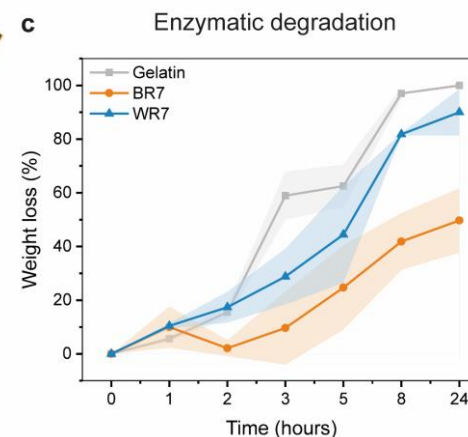


Fe³⁺ reducing capacity (pH 3.6)



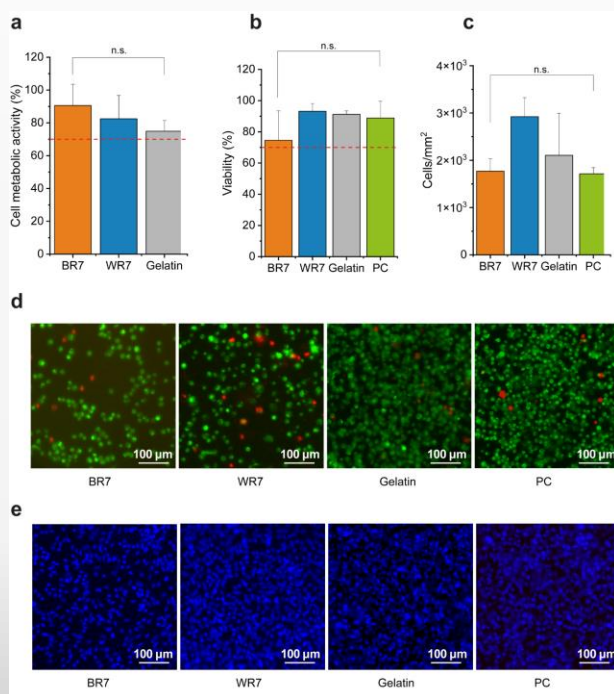
Proteolytic stability

Proteolytic stability was evaluated incubating the samples in a 0.01 U/mL protease XIV solution (pH 7.4), resulting in SDSY(3,5Br)GAP slowing down the degradation.



Functional properties

Biocompatibility

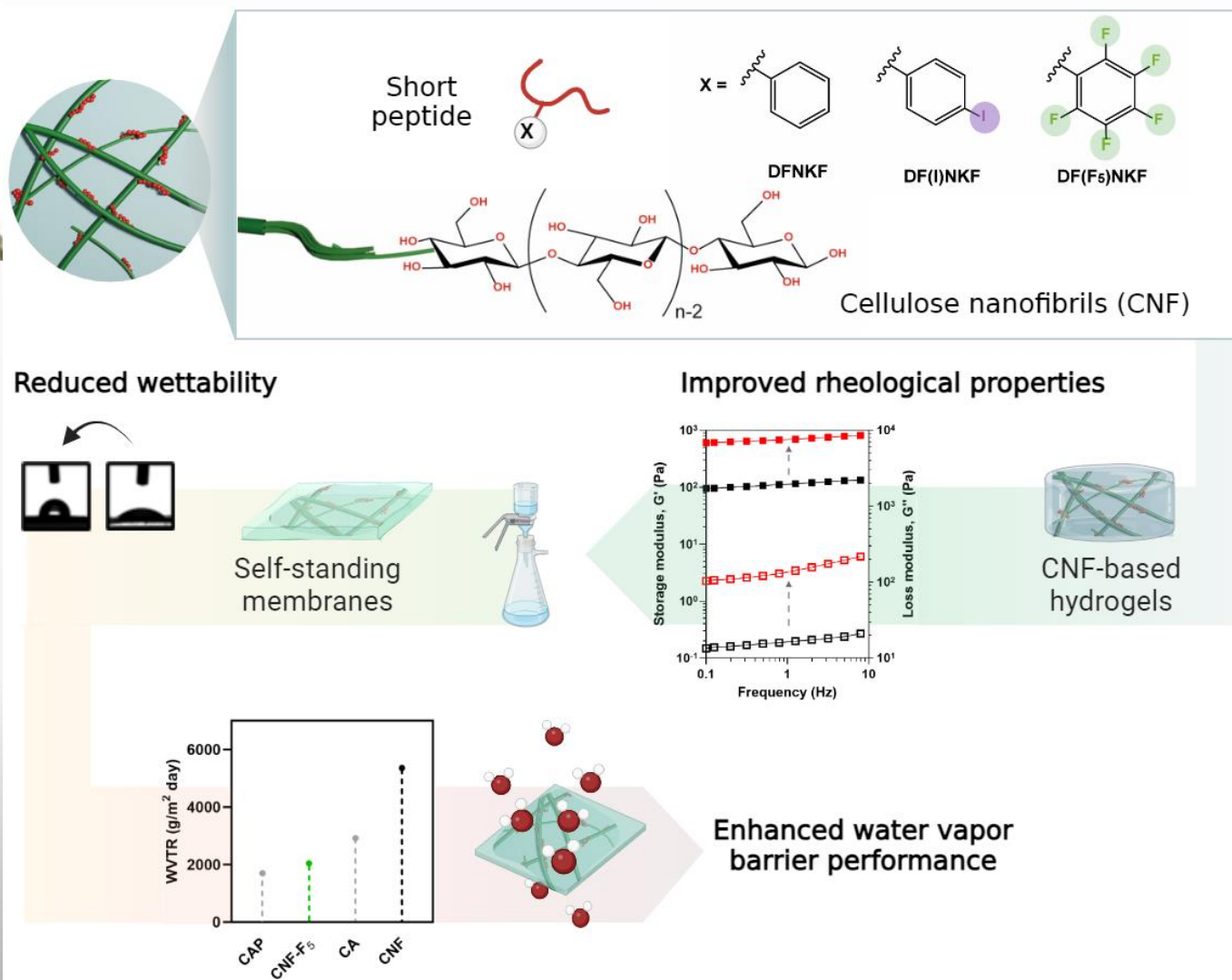


WR7 and BR7 scaffolds ensure a biocompatible environment suitable for endothelial cell culture and growing

Biological studies performed by Chiara Russo, DEIB, Politecnico di Milano



E. Marelli

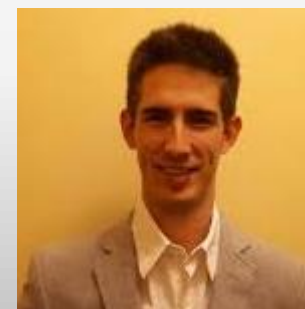




Acknowledgements



E. Marelli



A. Marchetti

A. Marchetti, E. Marelli, *et. al.*,
***J. Mater. Chem. B* 2024, 12, 9229.**



Keratin smARt mATerials from feather wasteE



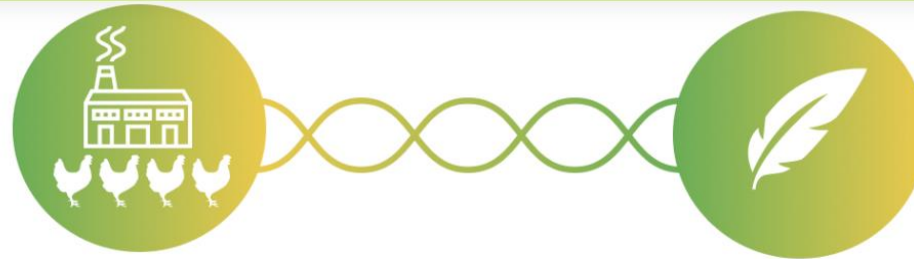
<https://www.karateproject.eu/>

Project

Partner

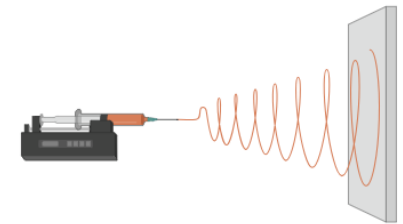
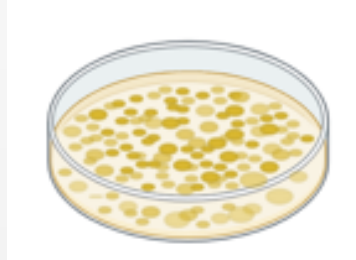
Advisory

Contact



global production of poultry meat
130 MILLION TONS

feather waste
3.1 MILLION TONS



Feather waste

**4 to 10 days incubation with
keratolytic bacteria strand**

**Keratin-materials:
Electrospinning**

Fondazione Cariplo is gratefully acknowledged for funding the KARATE project (2022-0437).

The global production of poultry meat is estimated to be ~130 million tons and, only in Europe, 13.4 million tons of poultry meat are produced annually, yielding 3.1 million tons of feather waste.



Conclusions



In the last two years, there has been an upsurge of examples demonstrating that using halogen bonding in supramolecular materials design is a vibrant new field of research.

- Each of the unique characteristics inherent to halogen bonding – directionality, tunable interaction strength, hydrophobicity, and large donor atom size – is shown to bring about or enhance functionalities that are absent without these characteristic features.
- Iodination of amyloidogenic peptides strongly promotes their fibril formation ability and affects the structure of formed fibrils.
- The strong effect of iodination on peptide self-assembly is demonstrated to be general and not limited to the core sequence reported here.
- Iodination of aromatic amino acids may thus develop as a general strategy for the design of new hydrogels starting from unprotected peptides and without the use of organic solvents.
- Single atom mutations of amyloidogenic core sequences by replacing hydrogen with iodine atoms greatly promote fibril formation and self-assembly. Methods promoting the formation of mature fibrils are advantageous for use, among others, as scaffolds for tissue engineering, controlled drug release, surgical reconstruction applications, microfluidic devices, biosensors, and bioswitches.



Acknowledgements



www.suprabionano.eu



@SupraBioNanoLab

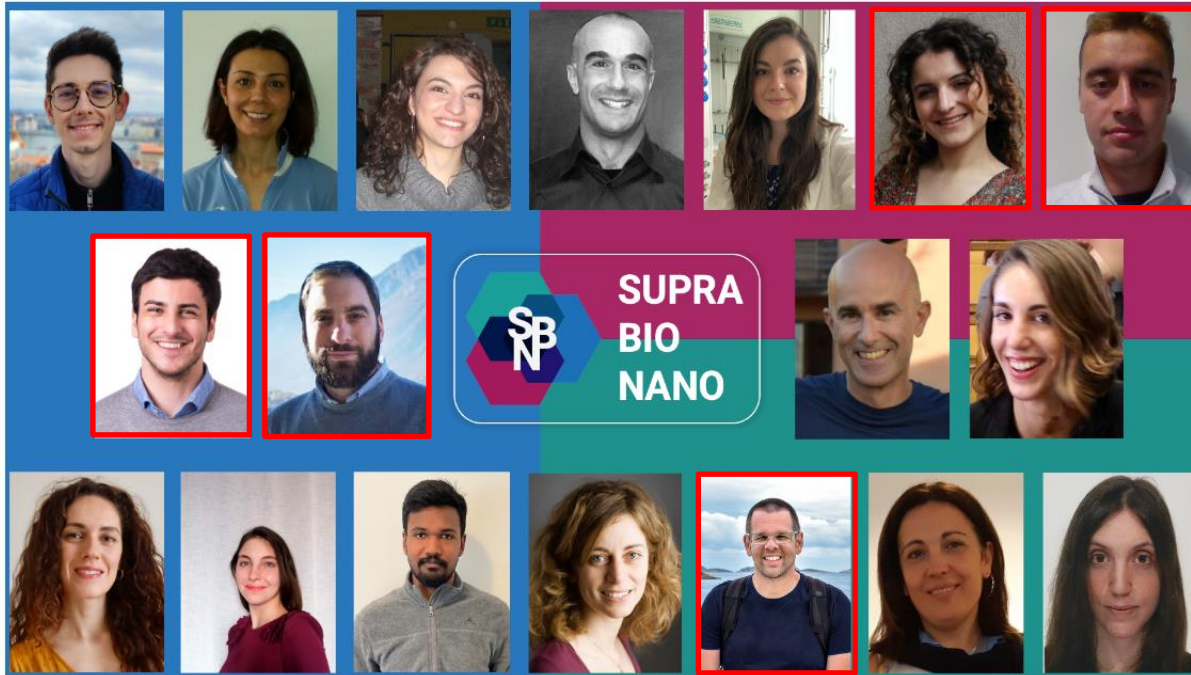


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The SupraBioNanoLab



Regione Lombardia



European
Commission



European Research Council

The project KARATE (grant no. 2022-0437) funded by Cariplo Foundation is gratefully acknowledged.



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